



Facoltà di Bioscienze e tecnologie  
agroalimentari e ambientali



# Application of nanostructured electrochemical sensor for food quality and safety

**Dario Compagnone**

Faculty of Biosciences for Food Agriculture and Environment

University of Teramo

# Food Safety and Quality



# Pesticides di Le uve

Trattamenti con dimetoato contro la mosca delle olive: cosa non ha funzionato?

E' l'anno di *Bactrocera oleae*, se ne parla in frantoio e in ogni dove, con scambi di consigli e opinioni. Tra miti di nuove **mosche delle olive** ogm e la realtà di dover cambiare le abitudini di una vita, anche in olivicoltura.

La Francia vieta l'importazione di ciliegie trattate con l'insetticida dimetoato (utilizzato l'anno scorso per 120 mila)

Prodotti fitosanitari: niente dimetoato contro la mosca del ciliegio

Sicurezza alimentare

Dimetoato, come è andata nella riunione tenutasi aprile scorso (avevano discusso) la discuss

## BOLOGNA

**Legambiente Emilia Romagna lancia l'allerta pesticidi "Troppe sostanze nelle acque e nelle città"**

Analisti denunciano la presenza di sostanze nocive messe al bando nel 60% delle analisi trovate fitosanitari

## Ambiente

**Greenpeace: quanti rischi per gli agricoltori dai pesticidi**

Le minacce per la salute dei lavoratori dei campi evidenziate in un rapporto dell'associazione ambientalista. Che ribadisce: serve un'agricoltura ecologicamente sostenibile

# (1) Q&S markers

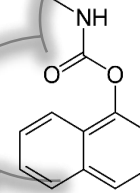
## Pesticides

- Control insects and other pests
- Application in terms of dose, timing and frequency
- Regulation (EC) No. 396/2005 and No. 1107/2009 of the European Parliament and the Council
- Maximum Residue Limit

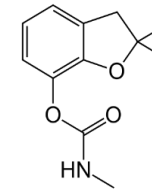


PLANTS

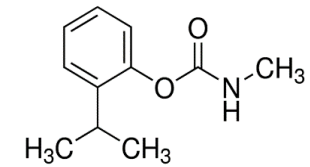
EU Pesticides database



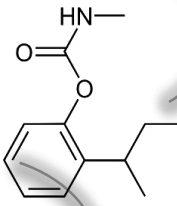
Carbaryl



Carbofuran



Isoprocarb



Fenobucarb



## (2) Q&S markers

Heterogeneous class of chemical compounds of considerable interest in the food industry:

### Sugars and Phenolic compounds...



Foods nutritional value/energy intake



Rheological properties  
(physical and structural attributes)



Consumer perception and sensory character:  
taste and flavour



key role in shelf-life (reducent ability,  
hygroscopicity/water control...)



Indicators of quality and ripening degree of  
horticultural product



Process indicators and substrate (e.g.  
fermentative process)



Nutritional property / functional food



Sensory characteristics definition  
(bitterness, astringency, etc...)



key role in shelf-life



Indicators of quality



Process indicators



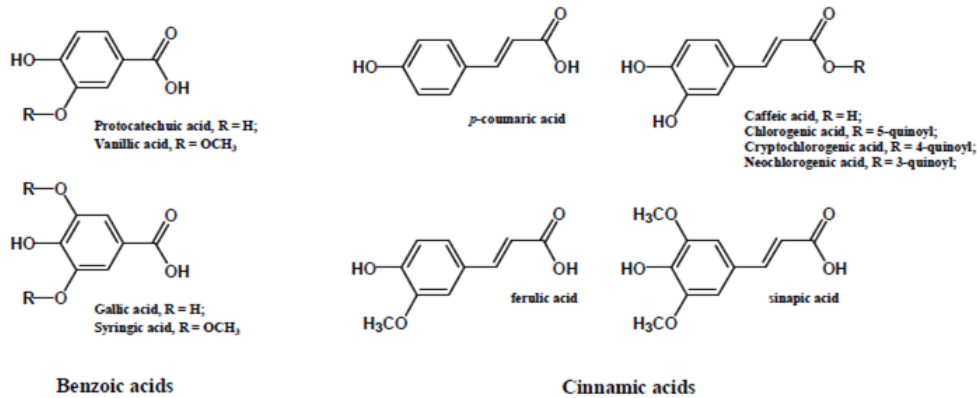
Anti-microbial property

# (3) Q&S markers

## Phenolic compounds in food

### Phenolic acids in food

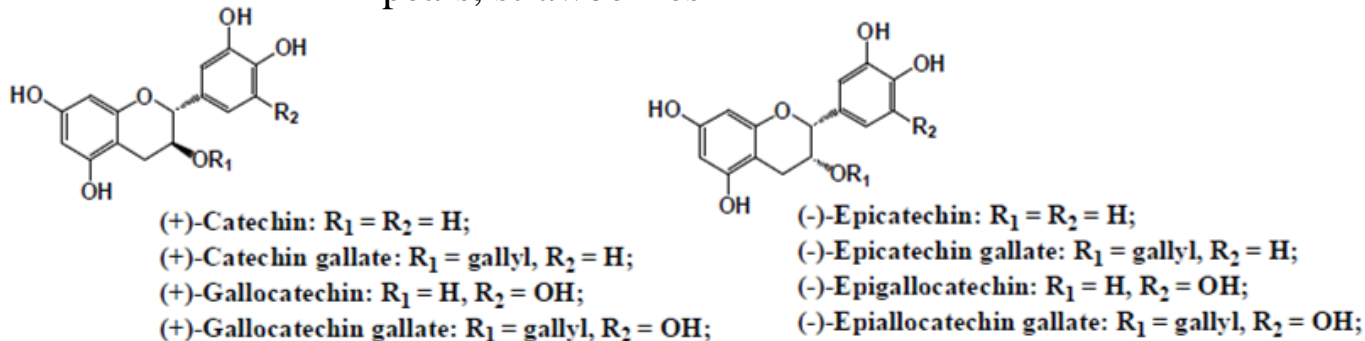
Fruits, vegetables, grains, seeds...



### Flavanols

Or flavan-3-ols or catechins.

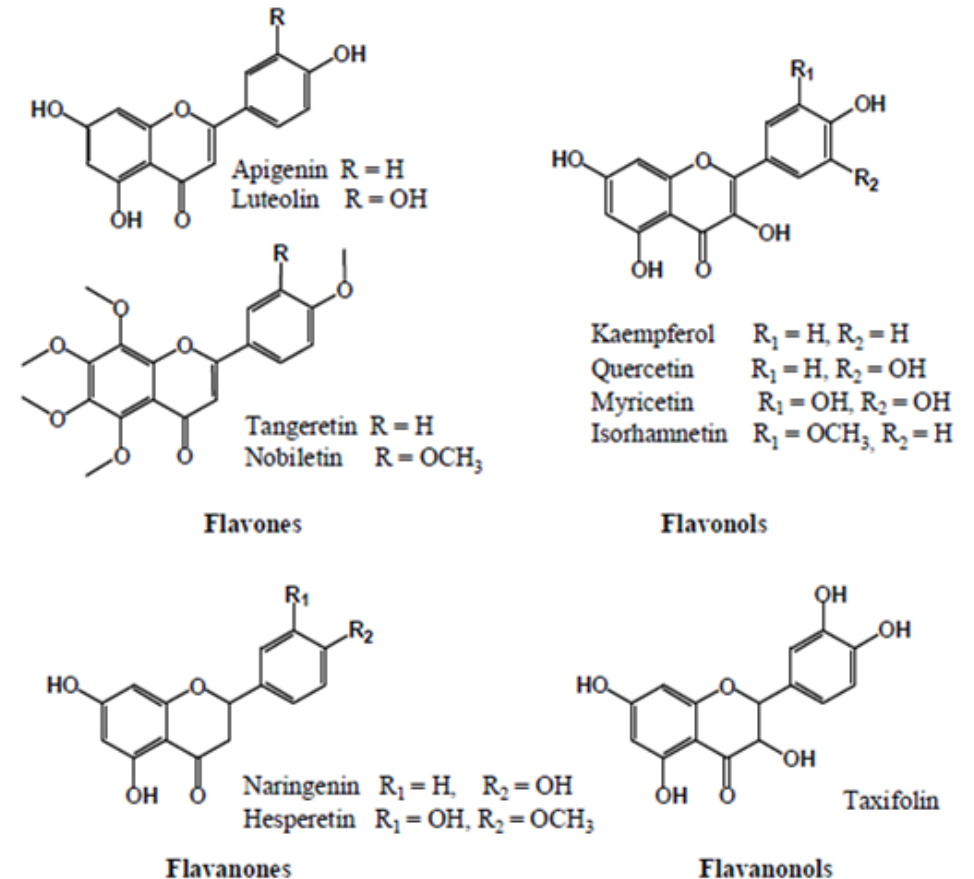
Black and green tea, apples, blueberries, peaches, pears, strawberries



### Flavonoids

Flavones, Flavonols, Flavanones and Flavanonols

Brightly coloured fruits and vegetables: blueberries, plums, apples, cherries, oranges, strawberries, spinach...

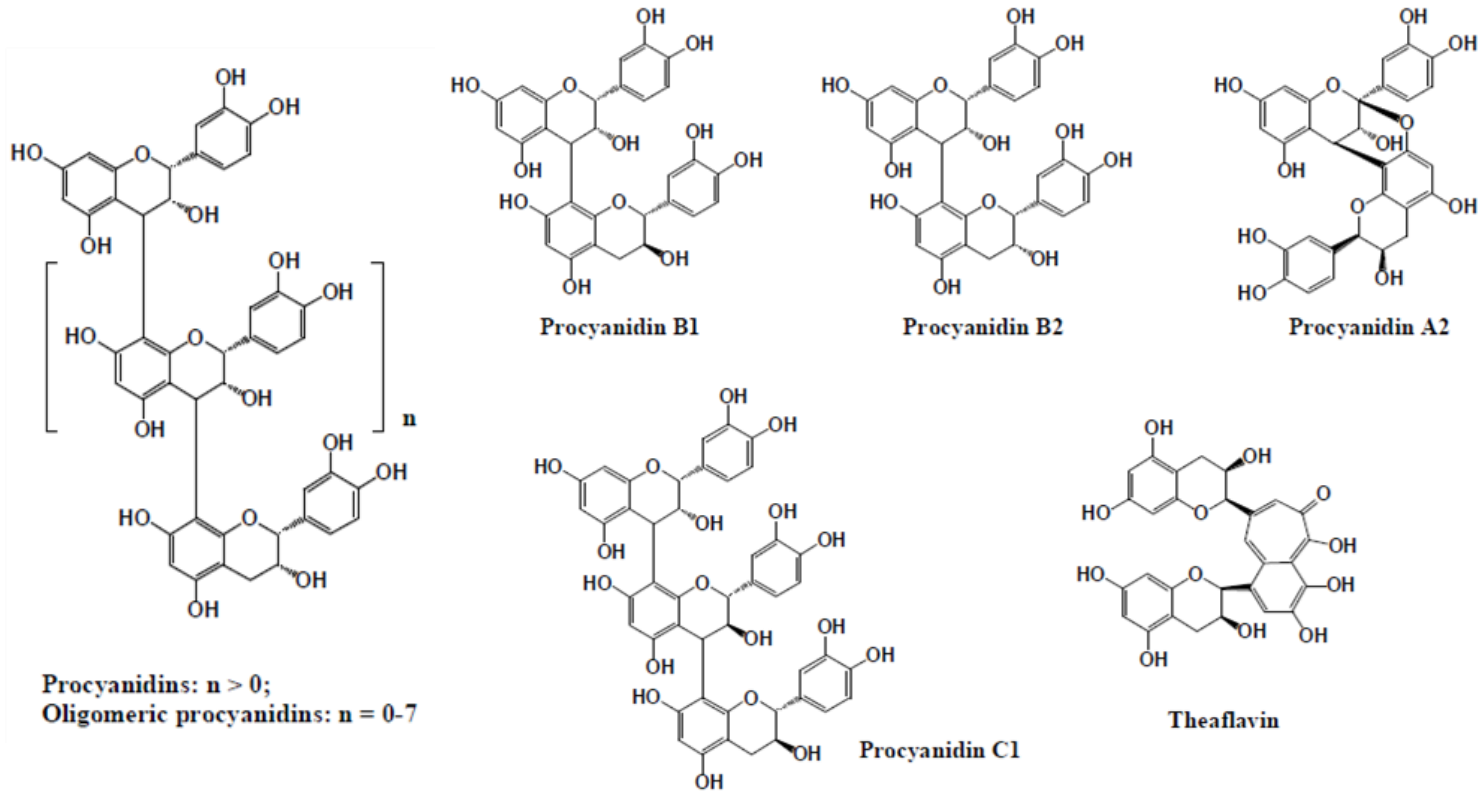


# (4) Q&S markers

## Phenolic compounds in food

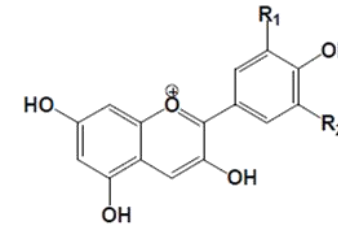
### Procyanidins

Grapes (seeds and skins), apples, chocolate and cocoa, red wines, blueberries, cranberries, pecans, pistachios



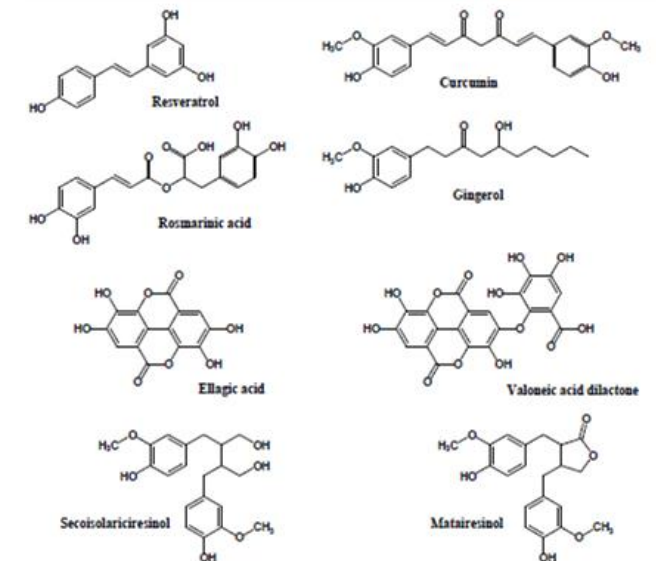
### Anthocyanidins

Blue and purple pigments food



Anthocyanidin	R <sub>1</sub>	R <sub>2</sub>
Cyanidin	-OH	-H
Delphinidin	-OH	-OH
Pelargonidin	-H	-H
Malvidin	-OCH <sub>3</sub>	-OCH <sub>3</sub>
Peonidin	-OCH <sub>3</sub>	-H
Petunidin	-OH	-OCH <sub>3</sub>

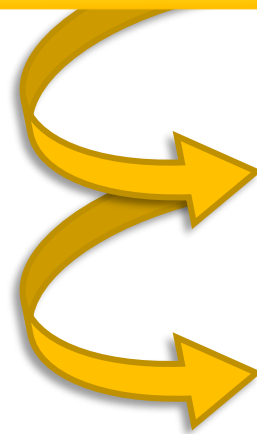
### Other important polyphenols...



# *HOW TO DETECT AND GUARANTEE FOOD Q&S?*



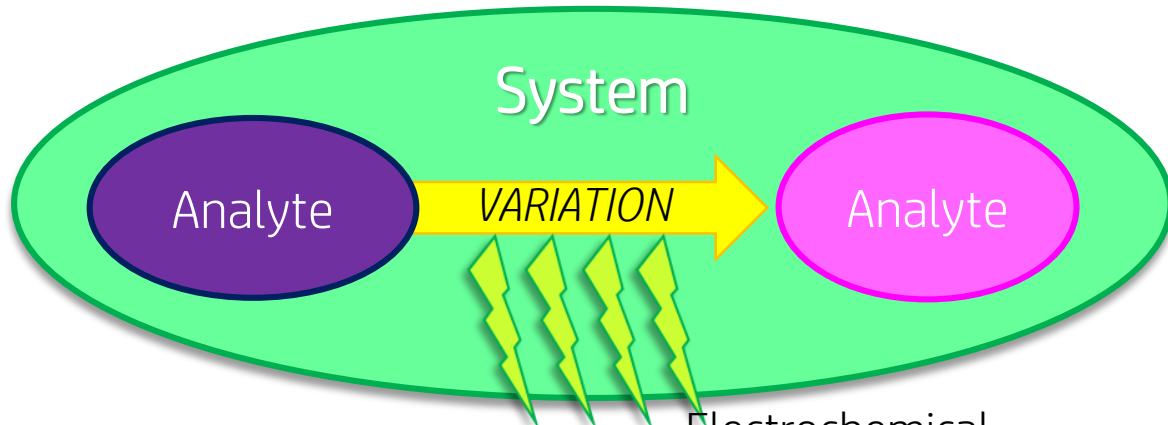
**Quality and Safety Markers**



**HOW TO DETECT ???**

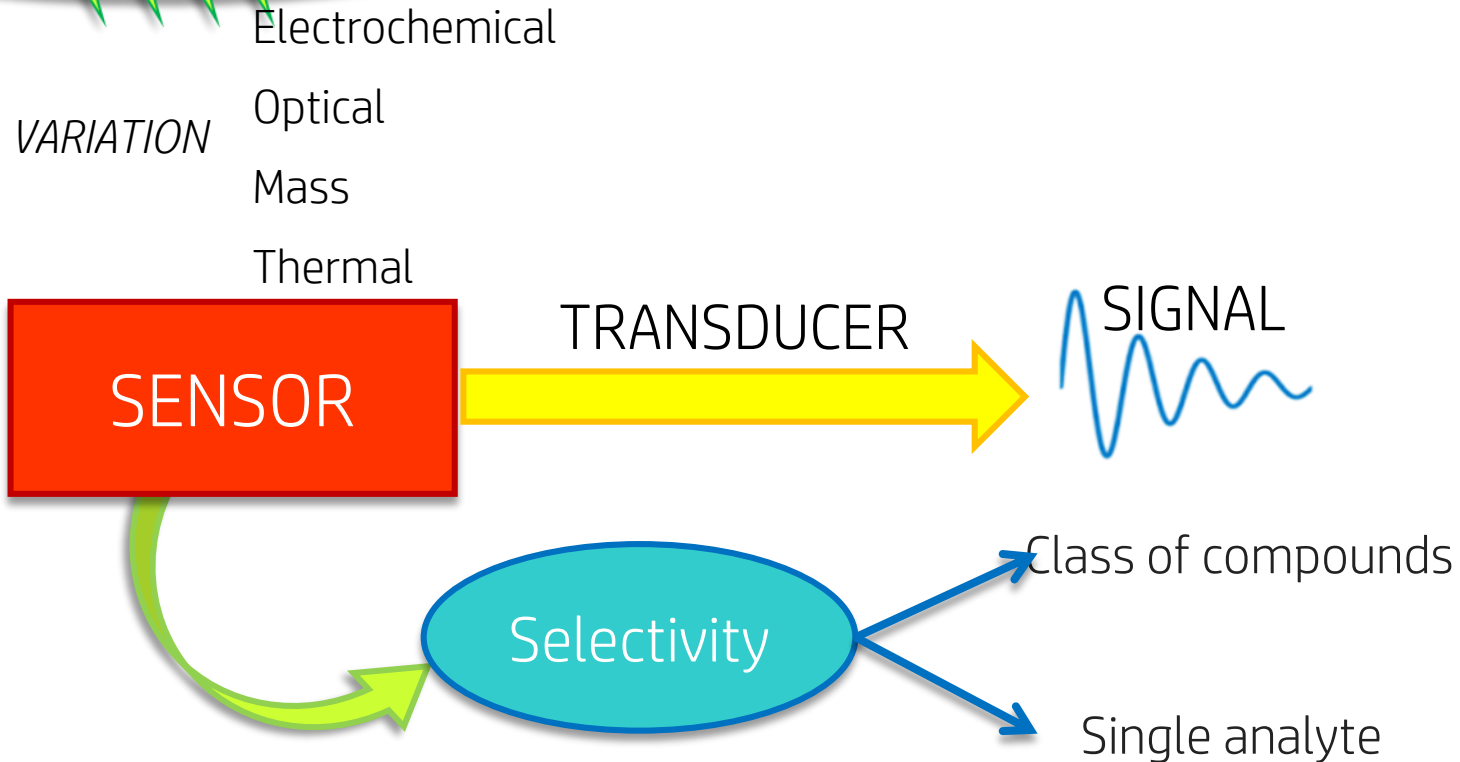
**SENSORS**



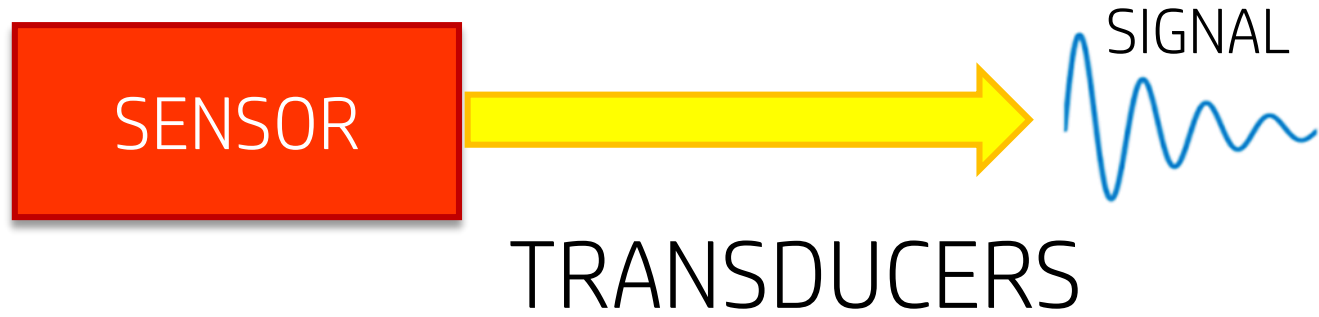


... is a device able to catch a physical, chemical or biochemical variation in a system...

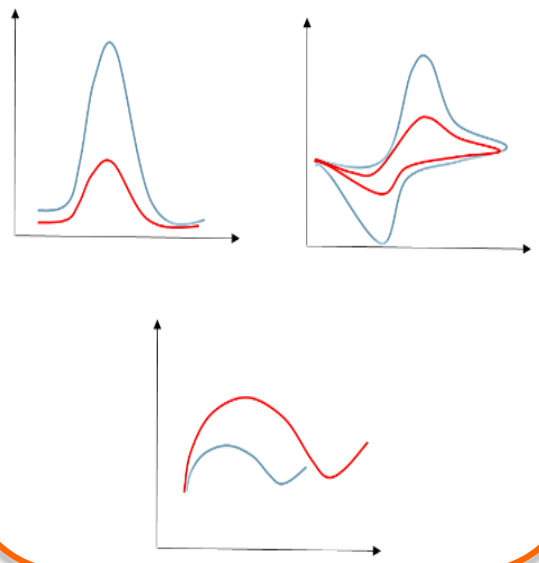
... the sensor devices response is able to give a qualitative and/or quantitative detectable response in an analytical way...



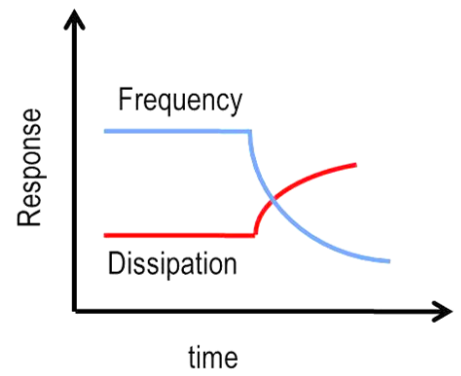
# (1) Sensors



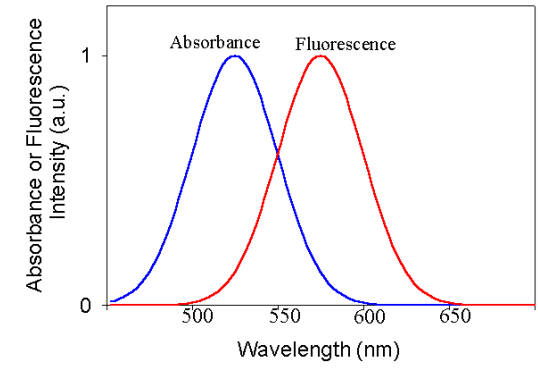
## Electrochemical devices



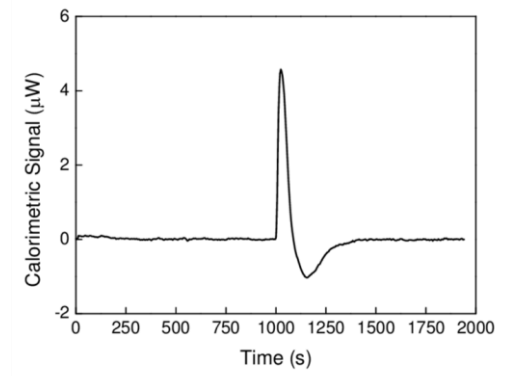
## Piezoelectric devices



## Optical devices

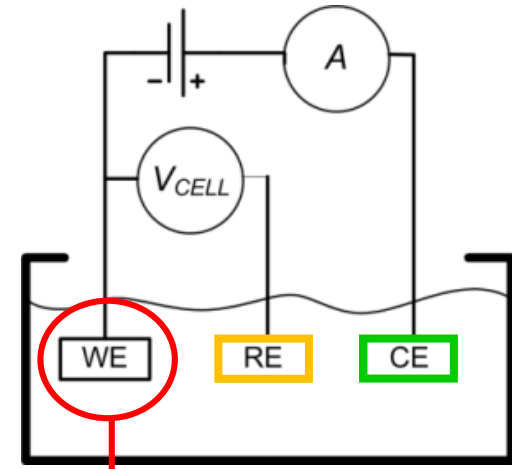


## Calorimetric devices

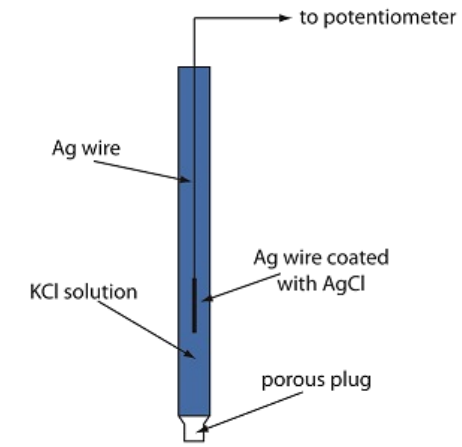


(3) *Sensor* → TRANSDUCER → *Electrochemical devices*

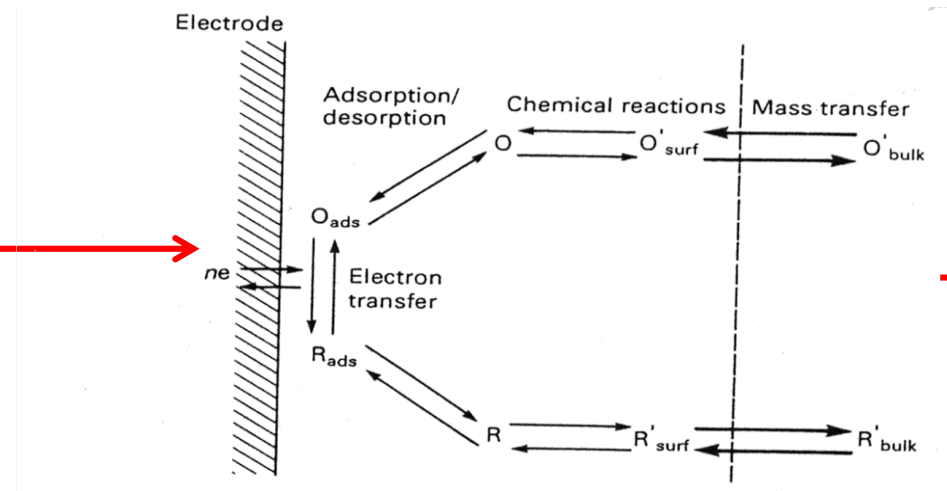
Working electrode (WE)



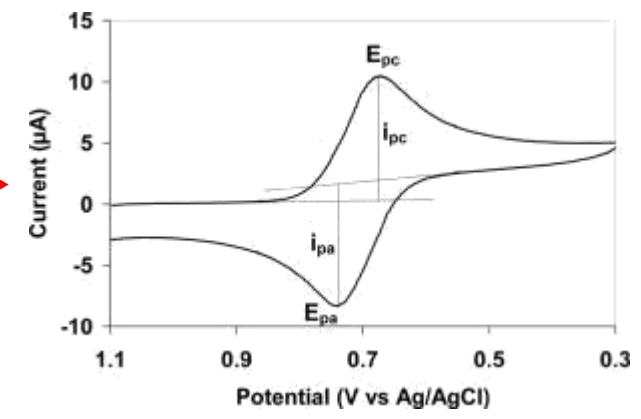
Reference electrode (RE)



Counterelectrode (CE)

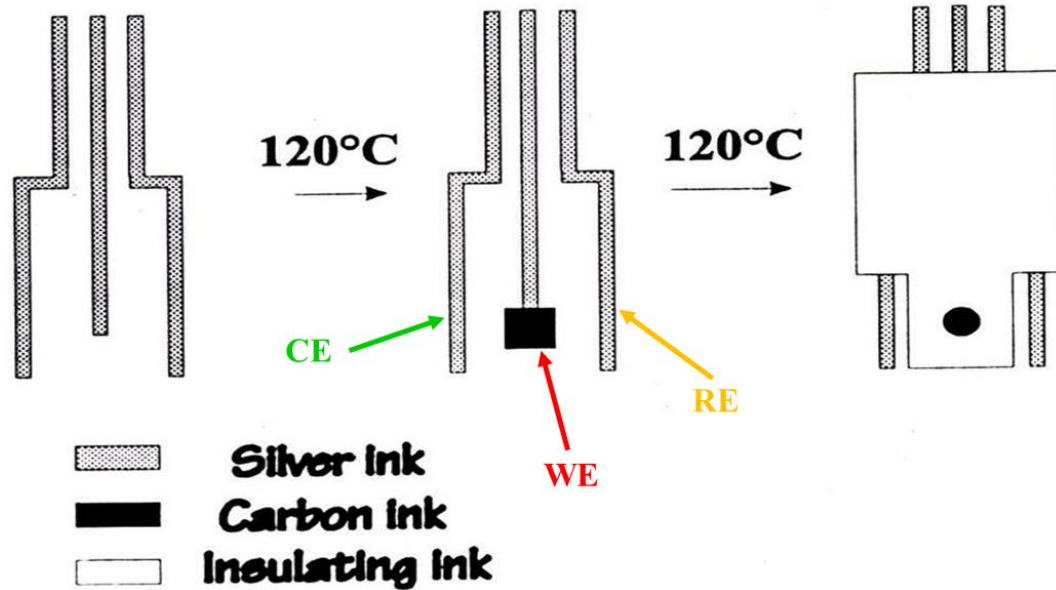
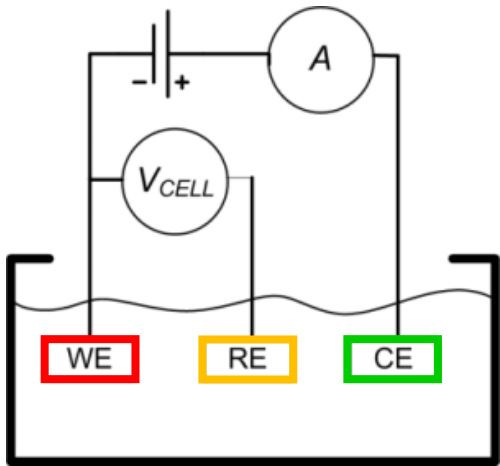


Pathway of surface processes involved in a general electrode reaction



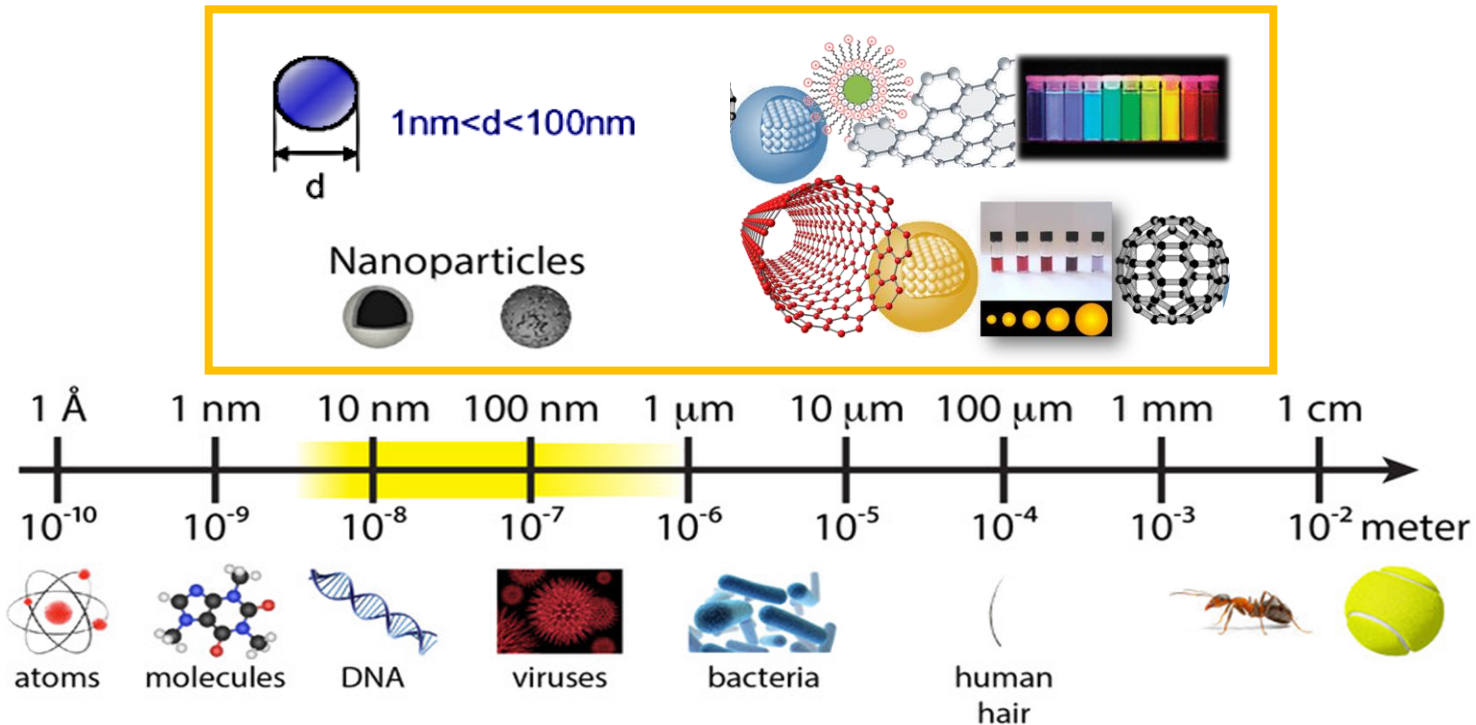
(3) *Sensor* → TRANSDUCER → *Electrochemical devices*

DISPOSABLE SCREEN-PRINTED CARBON ELECTRODES



- **Nanomaterial is defined by the European Commission Communication\* of 7 June 2005 like :**

‘means a natural, incidental or manufactured material containing particles, in an unbound state or as an aggregate or as an agglomerate and where, for 50 % or more of the particles in the number size distribution, one or more external dimensions is in the **size range 1 nm-100 nm**’

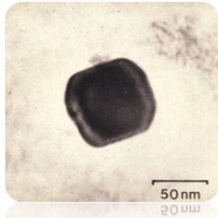


# (1) Nanomaterials

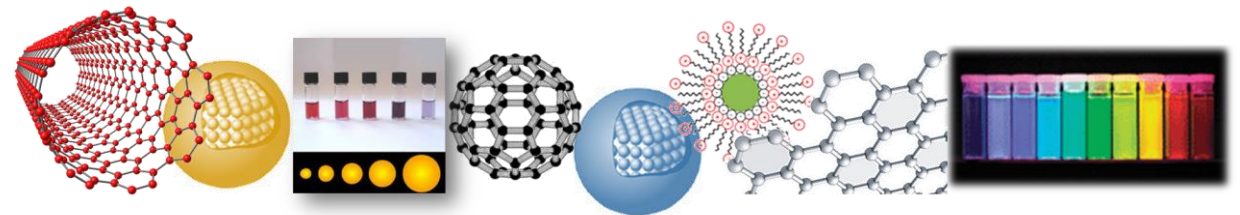
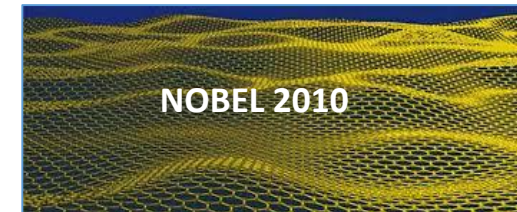
## Starting from Romans...

### Nanotecnologia romana

I colori cangianti della Coppa di Licurgo, datata IV secolo a.C., sono dovuti a nanoparticelle di oro e argento disperse nella matrice vetrosa.

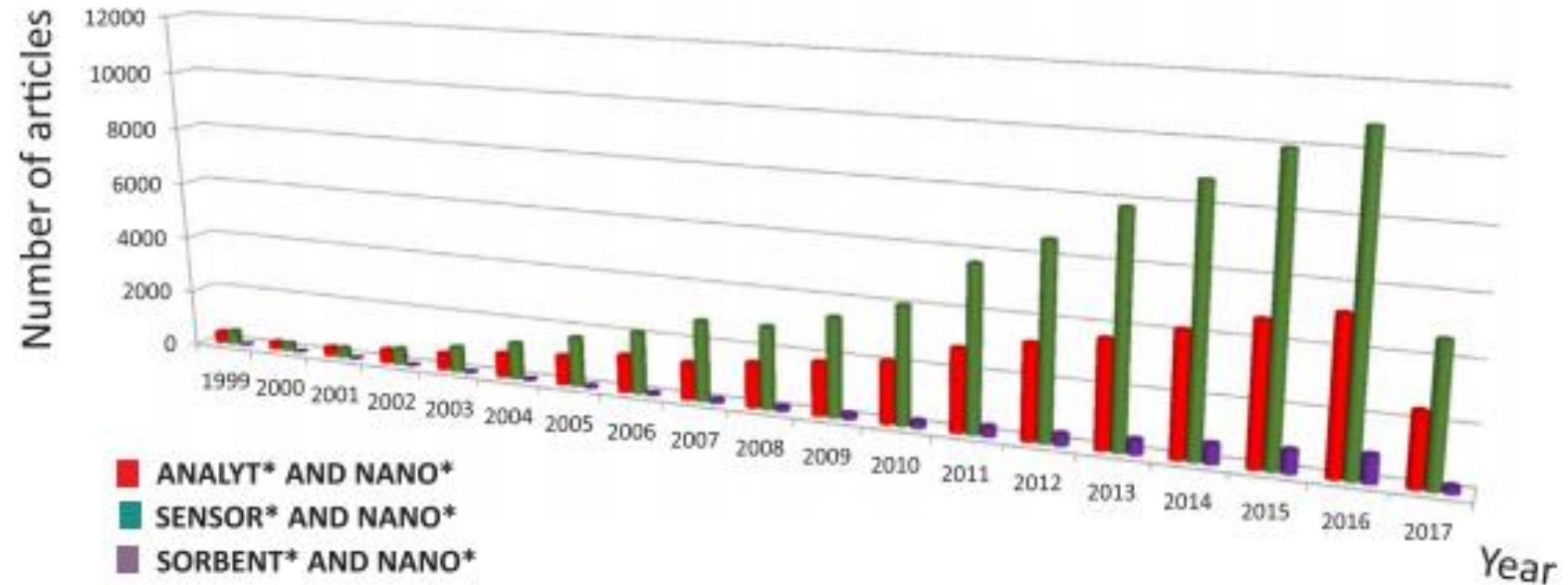
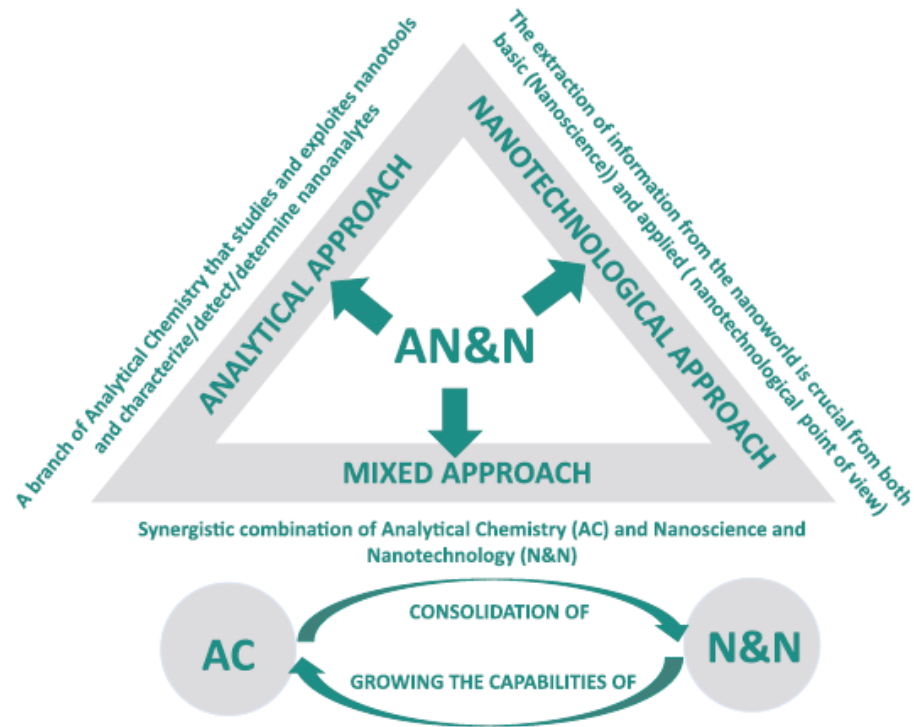


## ...till Today... looking towards the future...

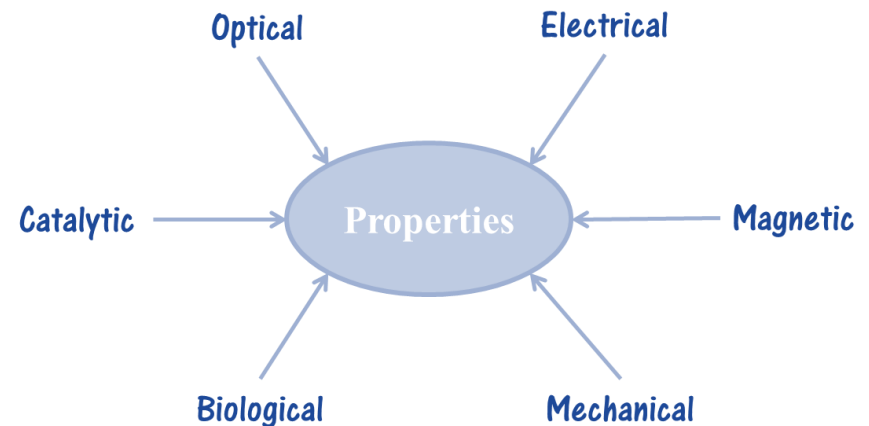


## (2) Nanomaterials

### Nanomaterials-based Sensors and sensing strategies



*Nanomaterials are widely used in various fields of analytical chemistry, because of their unique properties potential*



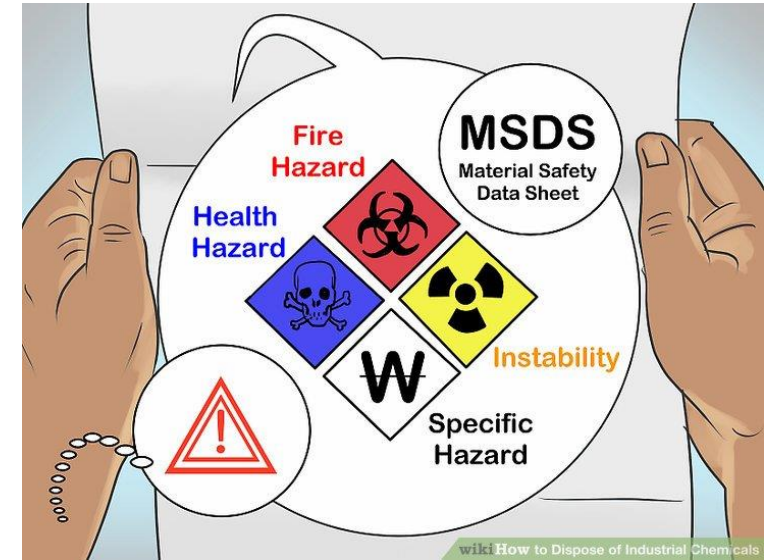
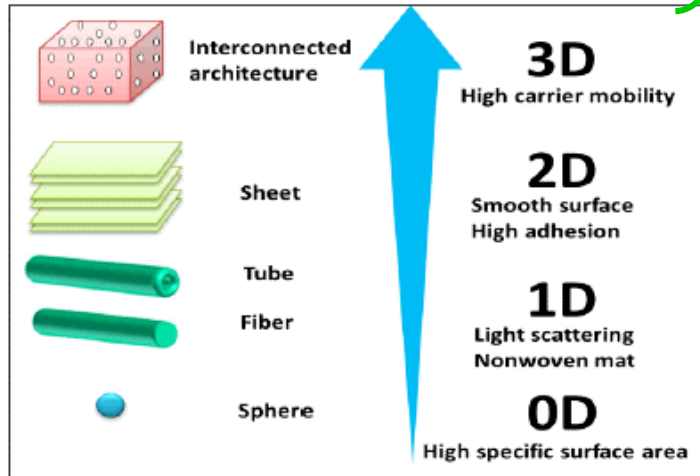
# (3) Nanomaterials

## Pros



- ✓ Different chemical nature
- ✓ Different morphologies
- ✓ High surface/volume ratio
- ✓ High functionalizability
- ✓ Easy interfaceability
- ✓ Size/morphology dependent properties → **tunability**

Sensor  
sensitivity & selectivity  
improvement



## Potential drawbacks

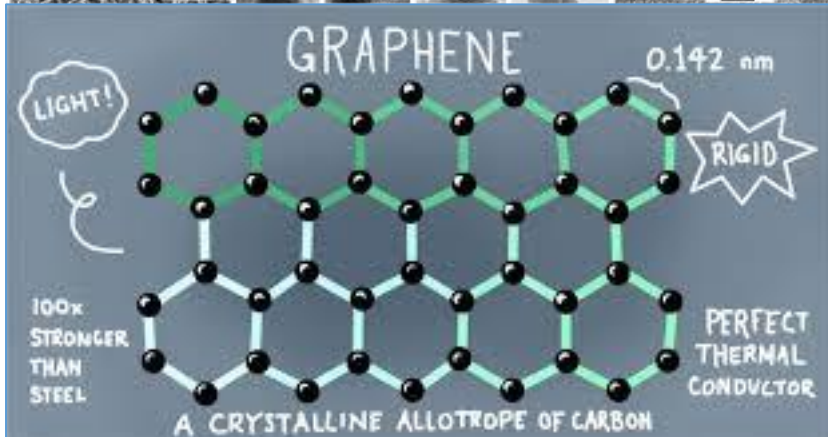
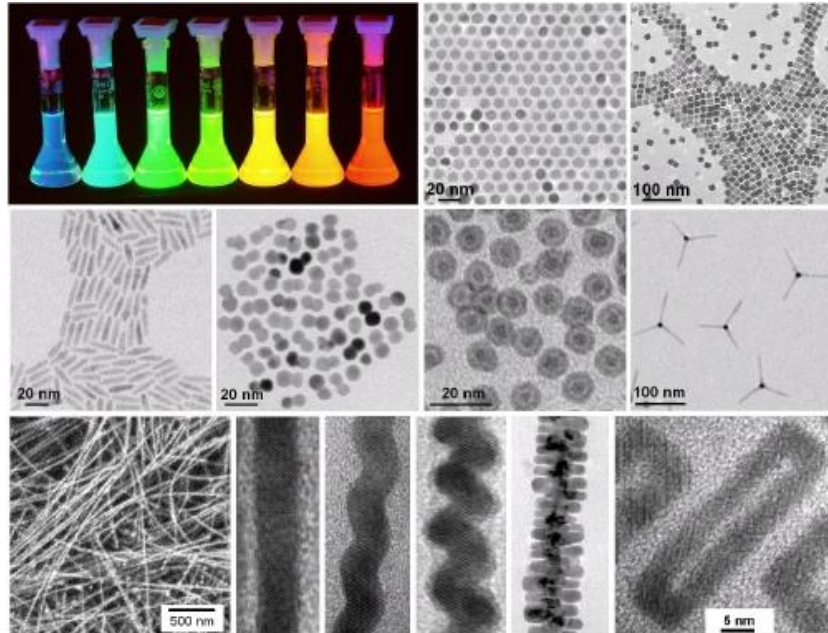


- ? Waste disposal
- ? Potential toxicity

Figure 1: Schematic illustration of structural dimensionality of nanomaterials with expected properties.



# Nanomaterials used in electrochemical sensors



## Carbon based nanomaterials:

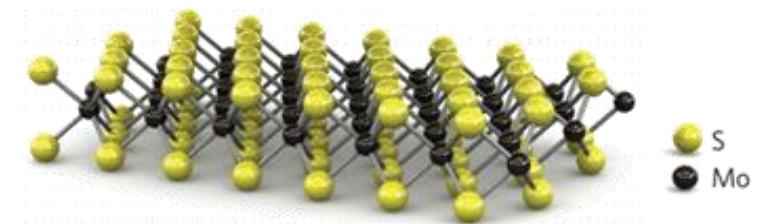
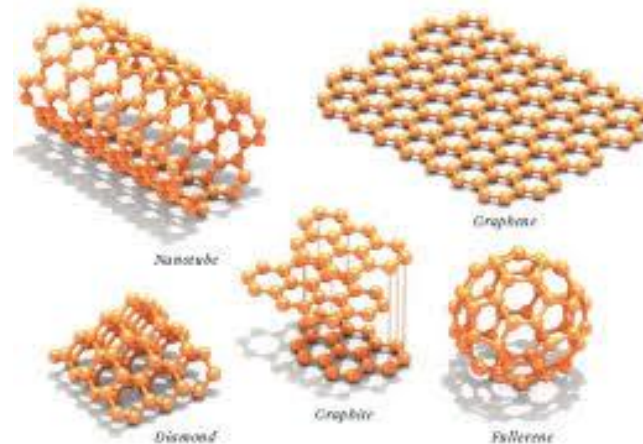
- Nanotubes
- Fullerenes
- Graphene
- Etc...

## Nanoparticles:

- Metal nanoparticles
- Metal Oxide nanoparticles

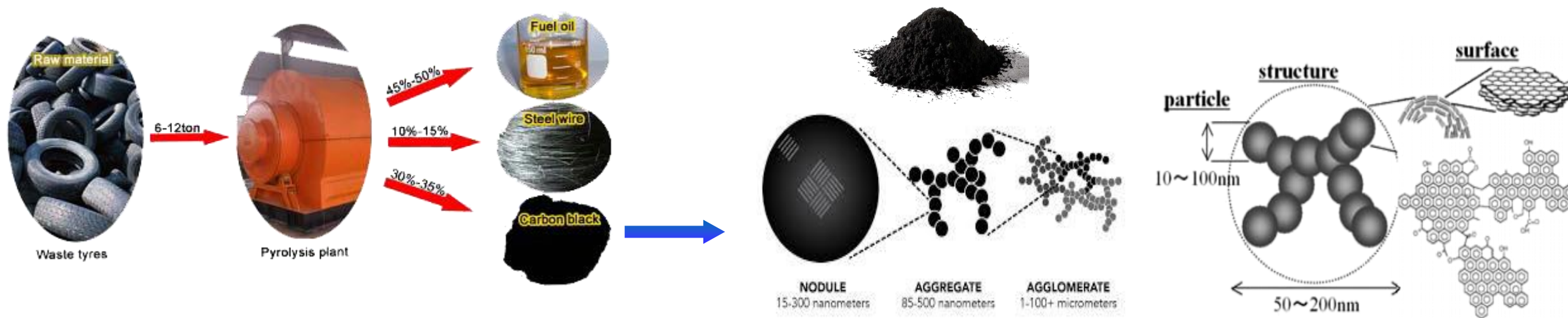
## Graphene-like nanomaterials:

- e.g. Transition Metal Dicalchogenides (TMD)



# Carbon based nanomaterials (*Carbon Black, CB*)

**Carbon black** is a material produced by the incomplete combustion of heavy petroleum products. It is mainly used as a reinforcing filler in tires and other rubber products. In plastics, paints, and inks, carbon black is used as a color pigment



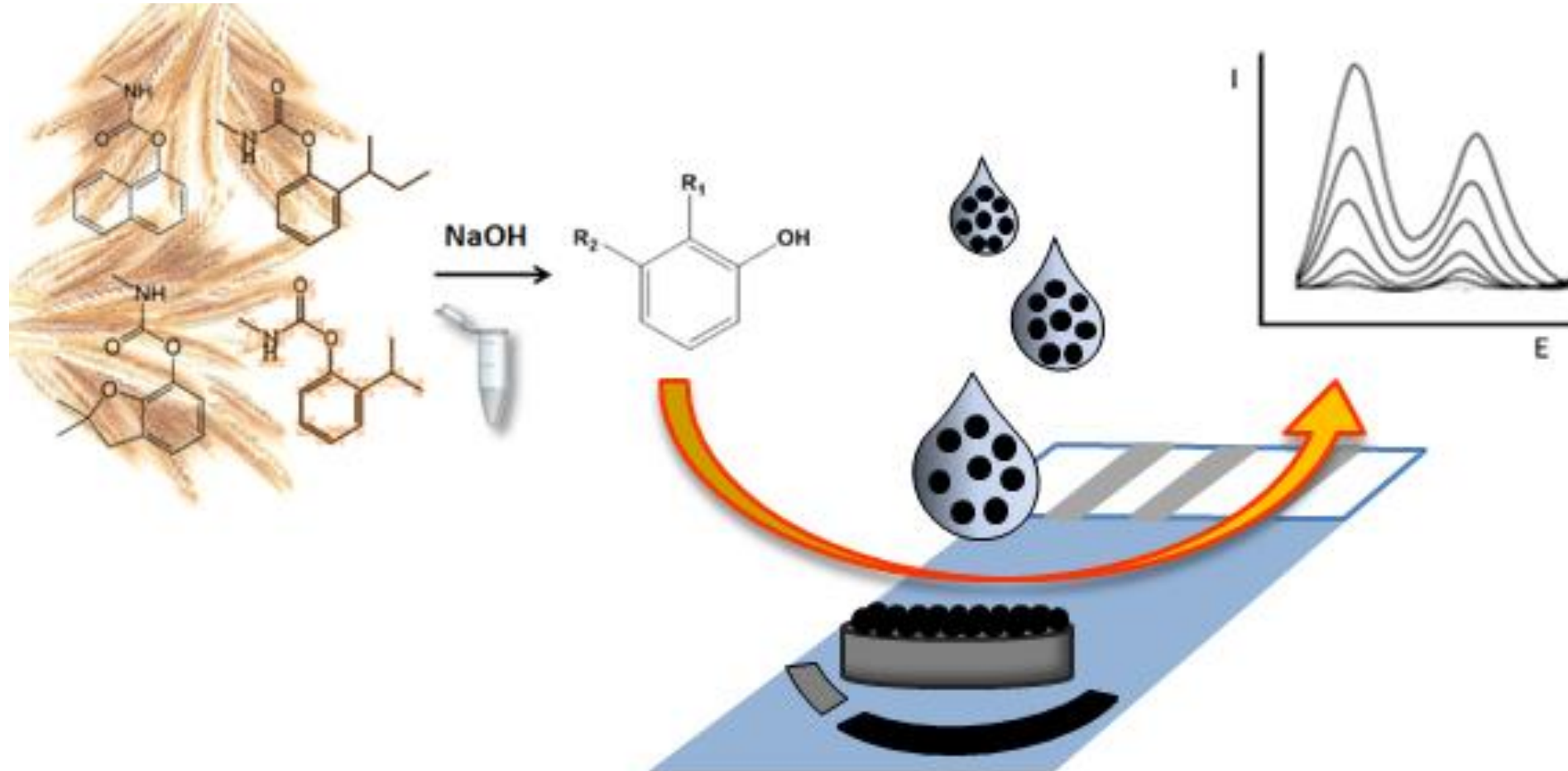
Electrocatalysis	➡	Selectivity
High surface	➡	Sensitivity
Resistance to fouling	➡	Reproducibility
Faster eletron transfer	➡	Improving separation performance

## ***CB compared with other nanomaterials:***

**Very low cost**  
**No synthesis**  
**No impurities due to synthesis**  
**Easily dispersible**  
**Large number of defect sites**

# (1) Carbon based nanomaterials (CB)

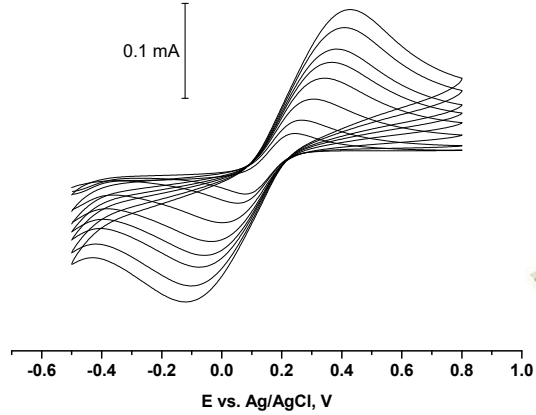
## SPE CBNPs for direct analysis of carbamates in grain samples



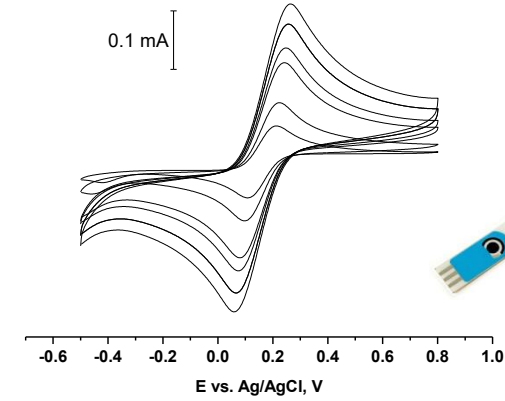
# (1) SPE CBNPs for direct analysis of carbamates in wheat

## SPE-CBNPs electrochemical behaviour for ferricyanide

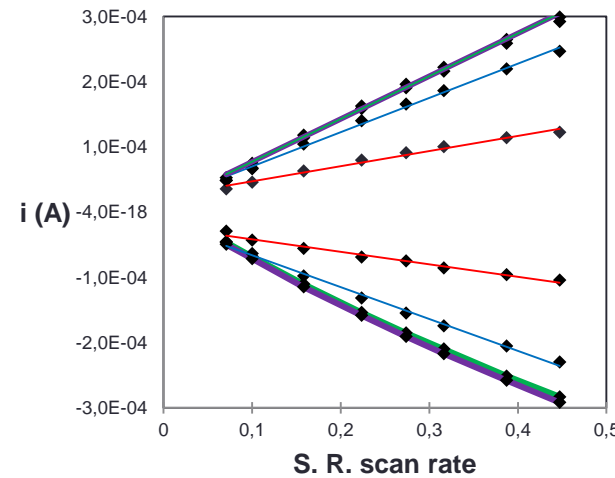
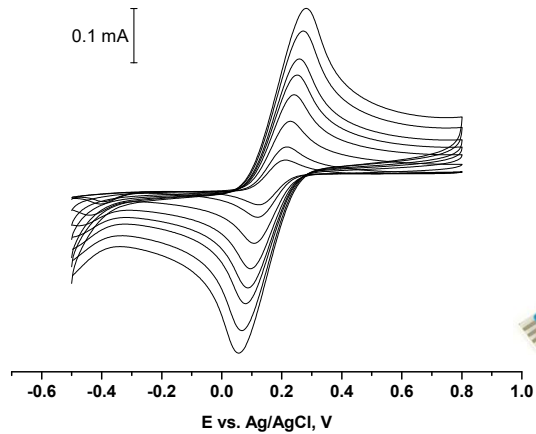
### Carbon SPE



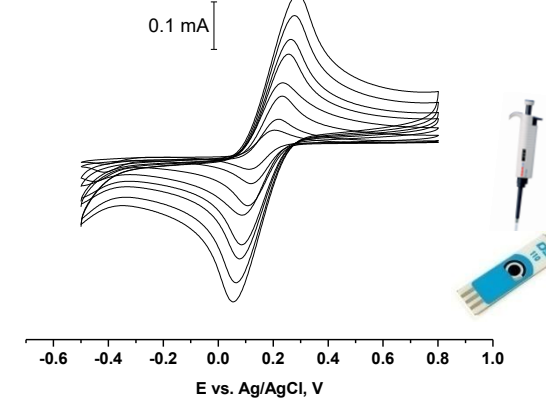
### Graphene/Carbon SPE (DS110-GPH)



### Carbon-CB SPE

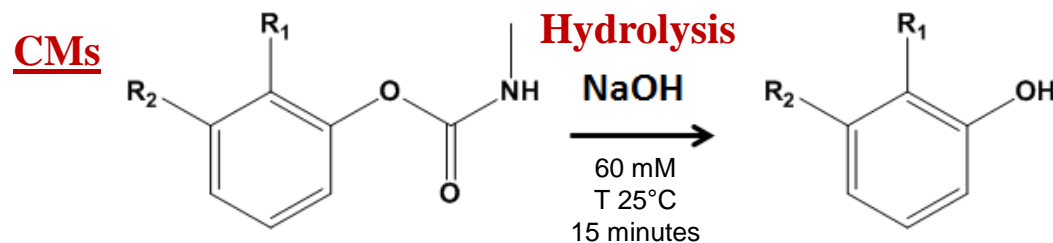


### Graphene/Carbon-CB SPE (DS110-GPH)



CB modified SPE demonstrates a better electron transfer

# (2) SPE CBNPs for direct analysis of carbamates in grain samples



SPE-CBNPs

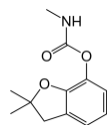
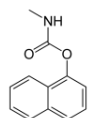
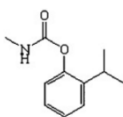
**DETECTION**

## SPE-CBNPs vs. CMs

**Isoprocarb**

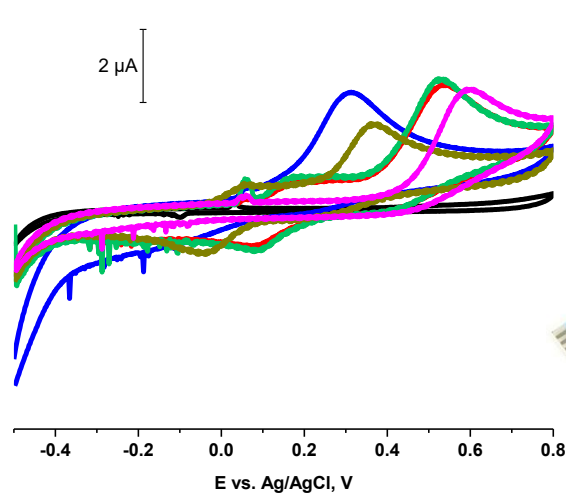
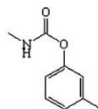
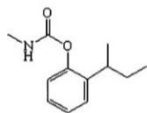
**Carbofuran**

**Carbaryl**

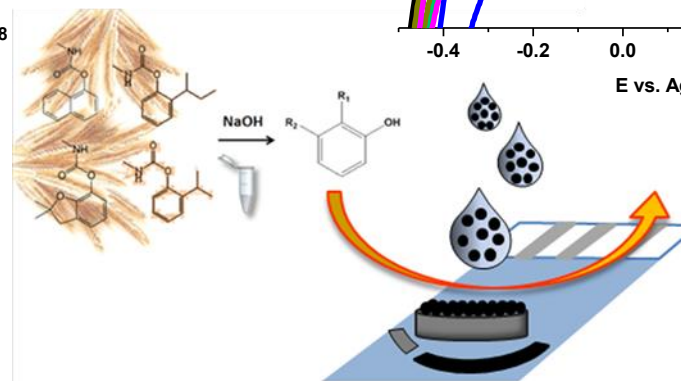
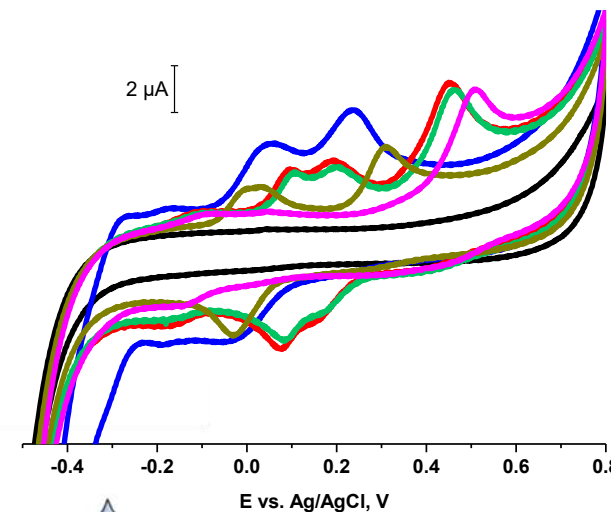


**Fenobucarb**

**Metolcarb**

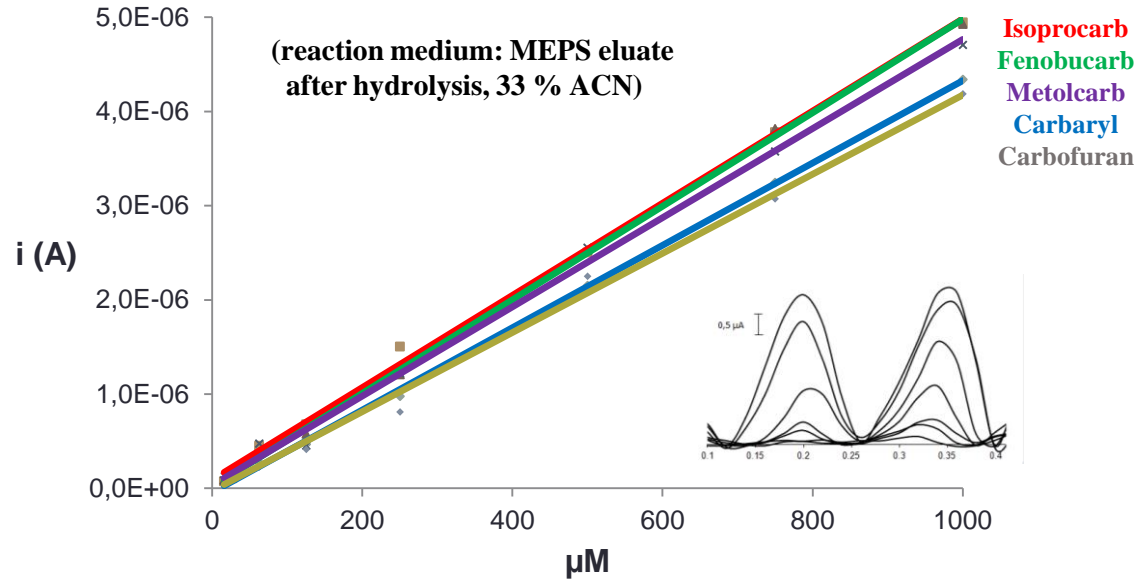


2 μA



# (3) SPE CBNPs for direct analysis of carbamates in grain samples

## SPE-CBNPs CMs Calibration, Reproducibility and Fouling resistance



**Peak intensity** (RSD, n=7): < 0.9 %

**Peak potential** (RSD, n =7): < 4,8 %

**Inter electrode reproducibility** (RSD, n=10): < 6.6 % p.i and < 3,4 % p.E.

**Fouling** (peaks RSD):

DPV (n = 30, 250 μM) 96 % v.s.32 %

CV (n = 20, 500 μM) 94 % v.s 15 %

Analyte	Linear range (μmol L <sup>-1</sup> )	Regression equation (Y=am + b)	Coefficient of determination (r <sup>2</sup> )	Detection limit (μmol L <sup>-1</sup> )	Quantification limit (μmol L <sup>-1</sup> )
<b>Isoprocarb</b>	0.1-100	y = 3E-08x + 5E-09	<b>0.9971</b>	0.6	<b>0.7</b>
Carbofuran	0.1-100	y = 6E-08x - 1E-08	<b>0.9999</b>	0.4	<b>0.5</b>
<b>Carbaryl</b>	0.1-100	y = 6E-08x + 2E-08	<b>0.9983</b>	0.4	<b>0.5</b>
<b>Fenobucarb</b>	0.1-100	y = 3E-08x - 8E-09	<b>0.9996</b>	0.6	<b>0.7</b>
<b>Metolcarb</b>	0.1-100	y = 6E-08x + 4E-08	<b>0.9980</b>	0.3	<b>0.4</b>



Nano carbon black-based screen printed sensor for carbofuran, isoprocarb, carbaryl and fenobucarb detection: application to grain samples

Flavio Della Pelle, Claudia Angelini, Manuel Sergi, Michele Del Carlo, Alessia Pepe, Dario Compagnone\*



# Nano carbon black-based screen printed sensor for carbofuran, isoprocarb, carbaryl and fenobucarb detection: application to grain samples (Hard wheat, organic hard wheat soft wheat, organic soft wheat, maize)

## Pesticide recoveries in grain samples

Recoveries : 78–102%

Correlation:  $r = 0.952$

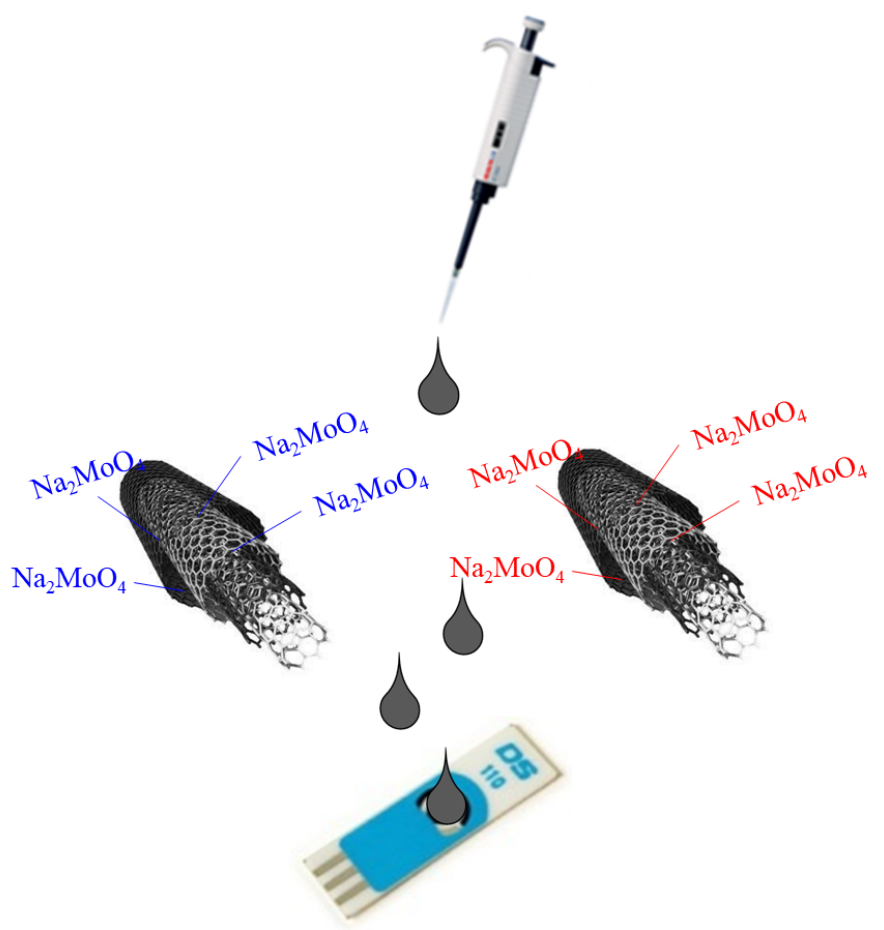
Accuracy: relative error between 9.0% and -7.8%

Analyte Spiked (mg Kg <sup>-1</sup> )	UHPL C-MS/MS recovery (%) <sup>a</sup>					CB-SPE recovery (%) <sup>a</sup>					Relative error (%)				
	HW	HWO	SW	SWO	MZ	HW	HWO	SW	SWO	MZ	HW	HWO	SW	SWO	MZ
<b>CA</b>															
0	< LOD	< LOD	< LOD	< LOD	< LOD	< LOD	< LOD	< LOD	< LOD	< LOD					
0.25	82 ± 6	87 ± 3	83 ± 9	89 ± 12	82 ± 15	88 ± 1	84 ± 4	76 ± 7	93 ± 8	85 ± 12	- 6.8	3.1	9.0	- 4.1	- 3.8
0.50	85 ± 2	84 ± 13	83 ± 2	81 ± 8	93 ± 7	88 ± 3	88 ± 11	80 ± 5	80 ± 6	90 ± 4	- 3.4	- 4.7	3.4	0.6	3.1
0.75	82 ± 10	78 ± 7	84 ± 1	82 ± 7	80 ± 4	85 ± 5	80 ± 9	87 ± 8	81 ± 9	84 ± 2	- 3.7	- 2.3	- 3.0	1.2	- 5.6
<b>CF</b>															
0	< LOD	< LOD	< LOD	< LOD	< LOD	< LOD	< LOD	< LOD	< LOD	< LOD					
0.25	80 ± 14	102 ± 10	83 ± 10	82 ± 14	81 ± 8	86 ± 9	97 ± 7	86 ± 6	79 ± 10	80 ± 9	- 7.8	4.8	- 3.7	3.4	1.9
0.50	78 ± 7	96 ± 6	78 ± 5	78 ± 5	79 ± 13	81 ± 4	100 ± 2	81 ± 3	80 ± 5	83 ± 11	- 3.3	- 4.8	- 3.6	- 2.9	- 5.6
0.75	79 ± 9	100 ± 9	84 ± 11	99 ± 131	79 ± 16	82 ± 5	100 ± 5	87 ± 7	100 ± 8	84 ± 9	- 3.4	- 0.6	- 3.1	- 0.7	- 6.4
<b>IC</b>															
0	< LOD	< LOD	< LOD	< LOD	< LOD	< LOD	< LOD	< LOD	< LOD	< LOD					
0.25	82 ± 5	80 ± 8	78 ± 8	84 ± 5	79 ± 8	82 ± 5	81 ± 6	82 ± 2	86 ± 9	80 ± 5	- 1.0	- 0.5	- 5.3	- 2.4	- 1.2
0.50	82 ± 7	85 ± 4	79 ± 2	78 ± 8	79 ± 5	81 ± 8	85 ± 4	81 ± 4	82 ± 6	78 ± 8	1.3	- 0.2	- 3.0	- 4.5	1.6
0.75	96 ± 2	96 ± 6	80 ± 3	95 ± 2	92 ± 14	96 ± 4	98 ± 9	80 ± 3	96 ± 4	97 ± 11	0.0	- 2.0	- 4.9	- 1.8	- 5.6
<b>FB</b>															
0	< LOD	< LOD	< LOD	< LOD	< LOD	< LOD	< LOD	< LOD	< LOD	< LOD					
0.25	83 ± 9	87 ± 7	83 ± 15	80 ± 4	87 ± 9	84 ± 5	92 ± 4	80 ± 13	79 ± 4	88 ± 3	- 0.9	- 5.3	8.0	4.5	- 0.9
0.50	79 ± 12	78 ± 9	97 ± 2	78 ± 5	80 ± 9	78 ± 8	82 ± 11	102 ± 0	78 ± 2	78 ± 6	7.7	- 5.7	- 5.2	4.7	5.1
0.75	89 ± 11	78 ± 8	102 ± 7	94 ± 13	79 ± 13	84 ± 11	81 ± 10	99 ± 3	96 ± 9	84 ± 11	5.6	- 4.1	3.3	- 2.5	- 6.8

<sup>a</sup> Mean value (n = 3) of three different extracts were employed for the recovery and relative error calculation for both CB-SPE and UHPLC-MS/MS methods.

# Carbon based nanomaterials (Carbon nanotubes, CNTs)

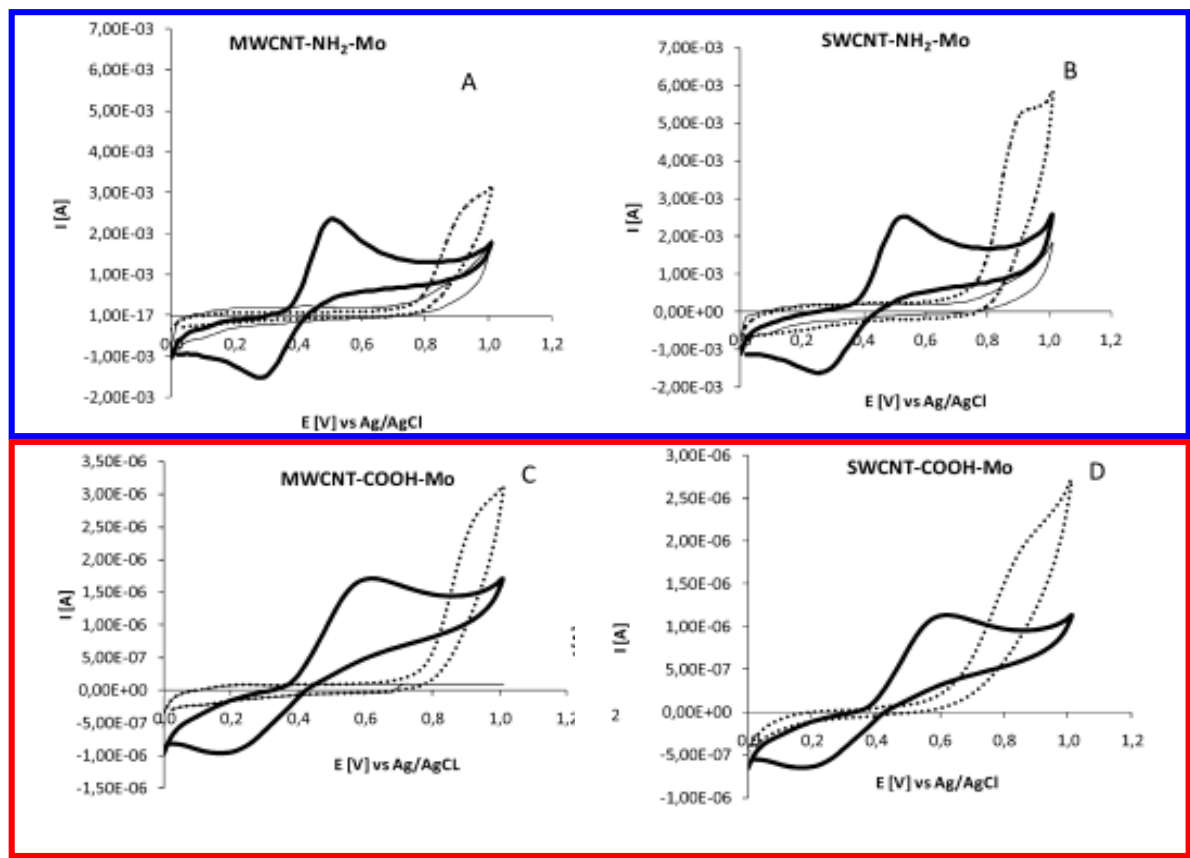
## Flow injection analysis (FIA) of olive oil polyphenols



-NH<sub>2</sub> bear  
 — Catechol  
 - - - Tyrosol

-COOH bear  
 — Catechol  
 - - - Tyrosol

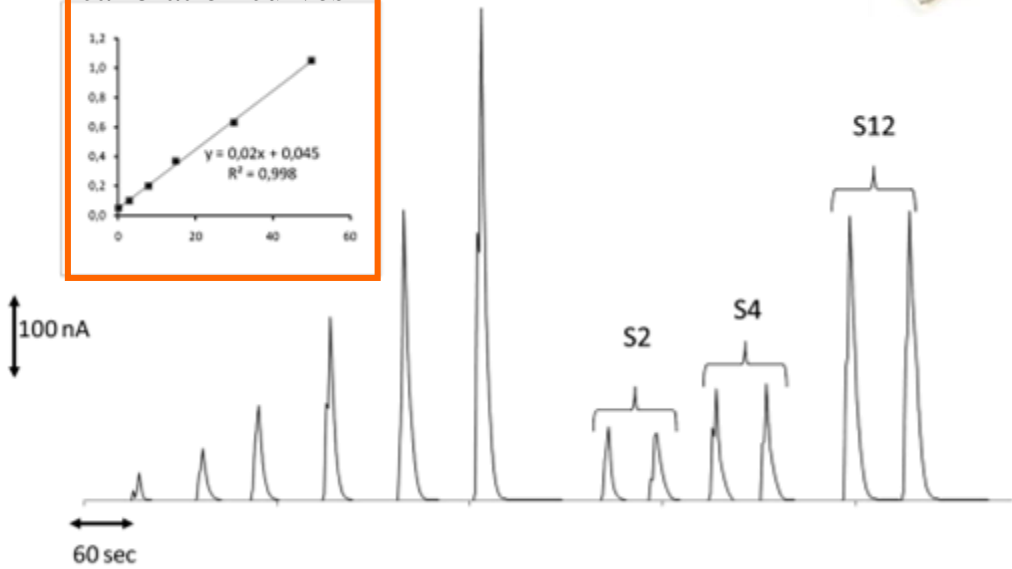
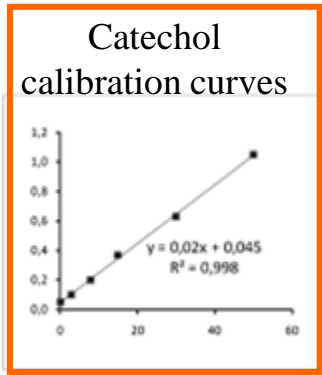
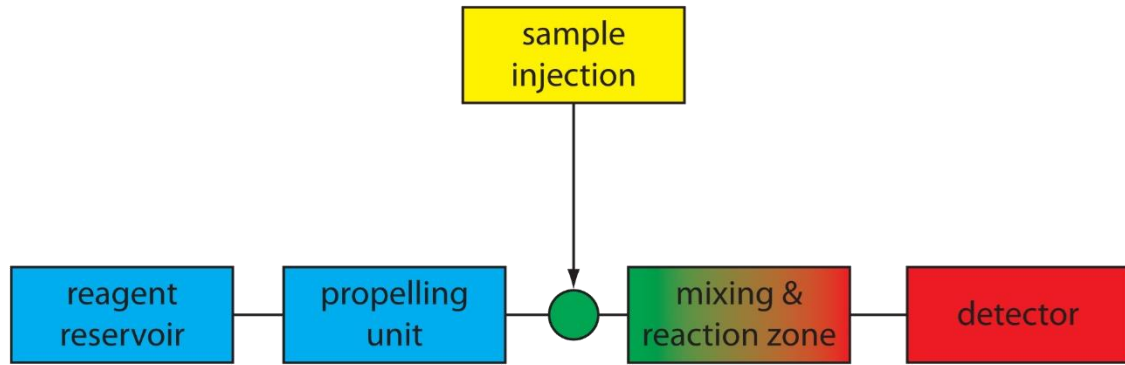
The -NH<sub>2</sub> functionalized CNTs exhibited also improved reversibility with respect to the -COOH functionalized CNTs



Comparison of the mediated electrochemical response of CNTs chemically modified nanosensor to catechol and tyrosol



# (1) Flow injection analysis



Amperometric FIA record of catechol standards and two replicates (S2, S4, and S12).

Full Paper

ELECTROANALYSIS

TOPICAL CLUSTER

## Selective Voltammetric Analysis of *o*-Diphenols from Olive Oil Using $\text{Na}_2\text{MoO}_4$ as Electrochemical Mediator

M. Del Carlo,<sup>a\*</sup> A. Amine,<sup>b</sup> M. Haddam,<sup>b</sup> F. della Pelle,<sup>a</sup> G. C. Fusella,<sup>a</sup> D. Compagnone<sup>a\*</sup>

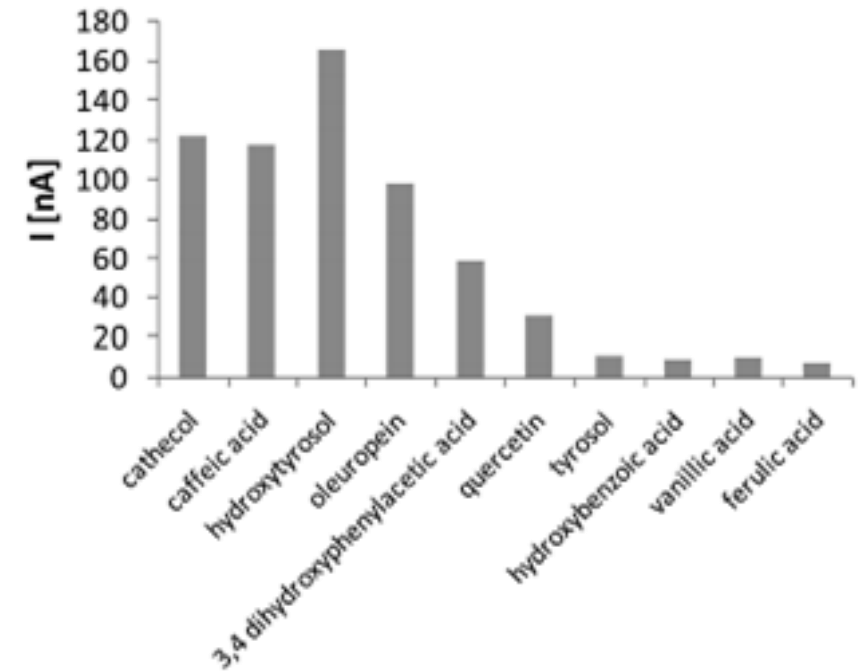
<sup>a</sup> Dipartimento di Scienze degli Alimenti, Università degli Studi di Teramo, Italia

<sup>b</sup> Faculté de Science, Université Hassan II, Mohammedia, Morocco

\*e-mail: mdelcarlo@unite.it; dcompagnone@unite.it

Received: October 20, 2011  
Accepted: December 24, 2011

Selectivity of the FIA amperometry method using a potential of + 380 mV; standard concentration was  $20 \mu\text{mol L}^{-1}$

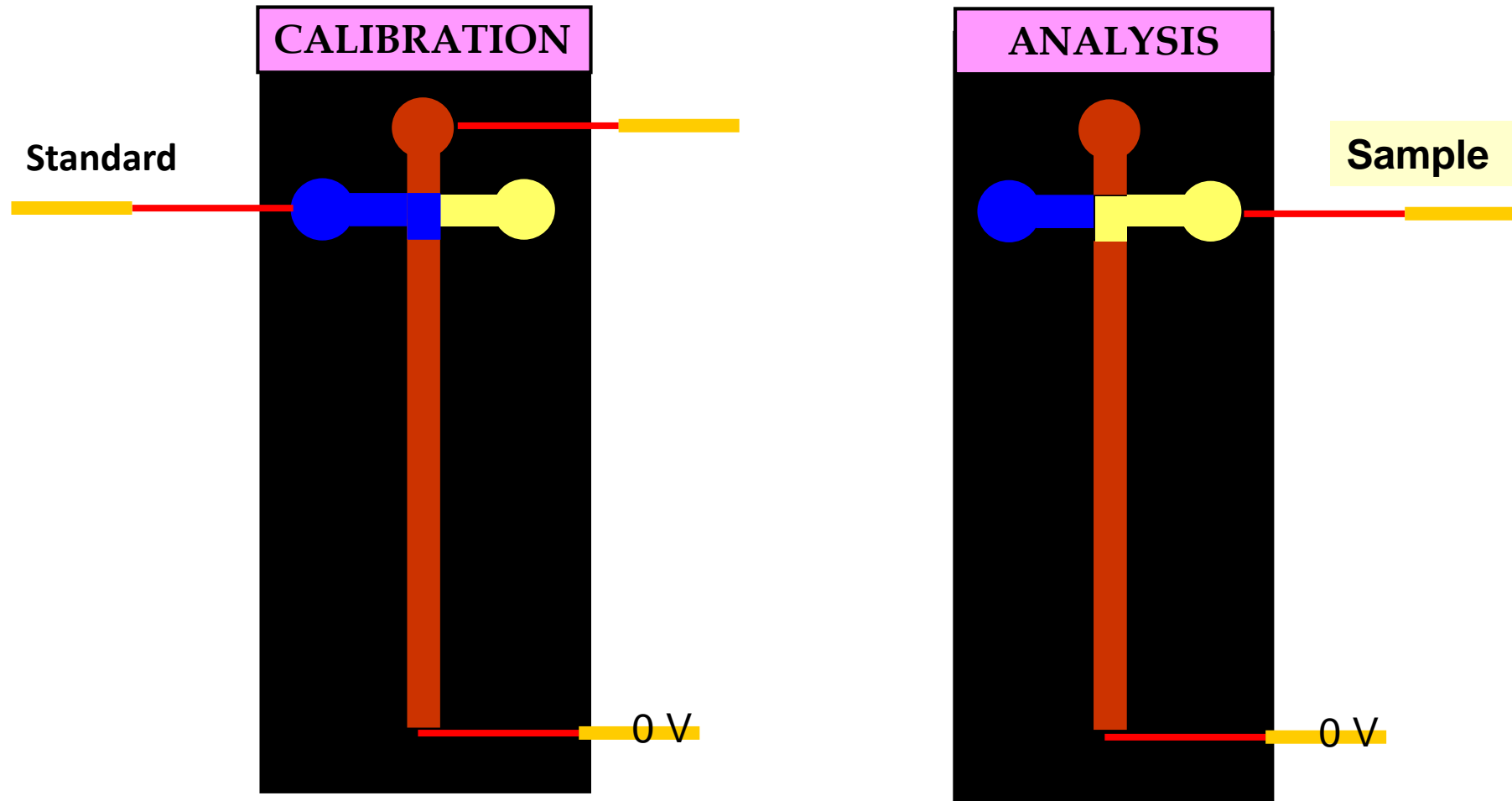


# Microfluidic LAB on CHIP strategy

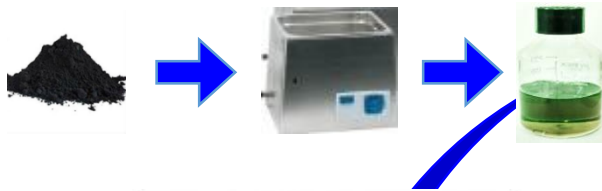
Why CE microchips?	Why Electrochemical Detection?
<p>High speed of separation <b>(seconds to few minutes!)</b></p> <p>High efficiency &amp; sample processing Low sample/reagent consumption <b>(from nL to pL!)</b> Extremely low waste generation <b>(“enviromental friendly”)</b></p> <p>Portability <b>(point of care testing)</b></p> <p>Disposability</p> <p>Highly multiplexed systems “Lab-on-a-chip”</p>	<p><b>Inherent miniaturization</b> without lossing of performance</p> <p><b>High compatibility with microfabrication technologies (easy to integrate)</b></p> <p><b>High compatibility with nanotechnologies (“easy “to integrate, and creativity!)</b></p> <p><b>High sensitivity</b> (some times approaching to fluorescence)</p> <p>Independence of optical path lenght and sample turbidity</p> <p><b>Not expensive!</b></p>

# Microfluidic LAB on CHIP strategy

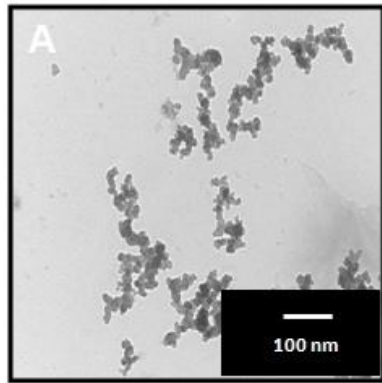
Microfluidic chips for reliable analytical determinations  
integrating a simplified calibration protocol



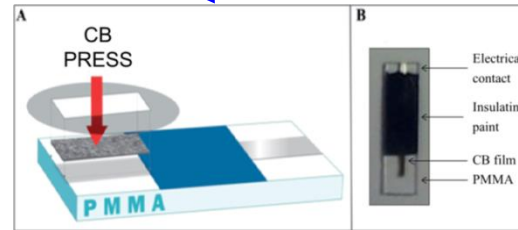
# Microfluidic LAB on CHIP strategy



## Is CB press transferable?



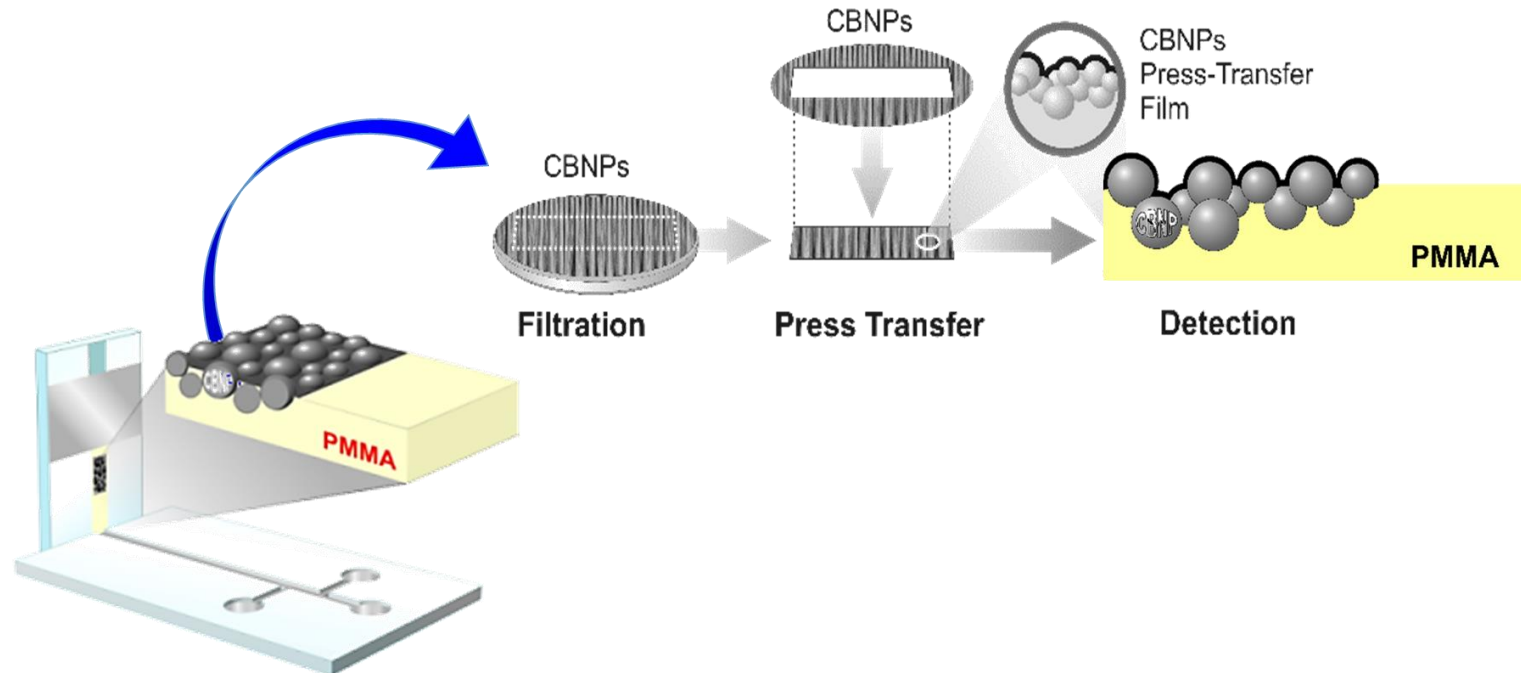
• CBNPs  $\varnothing \leq 30$  nm



Few nanomaterials can be transferred to pressure (SWCNTs) on plastic supports !!



## Fabrication strategy

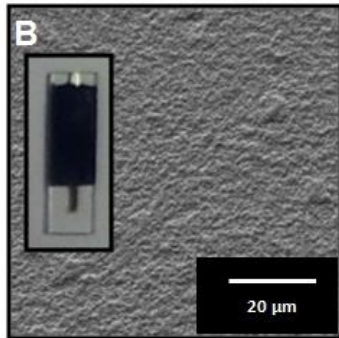


# (5) Microfluidic LAB on CHIP strategy

## CBNPs Press-transferred transducer characterization

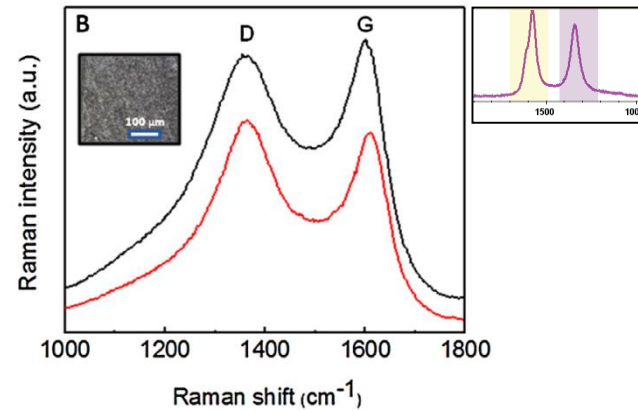
### FE-SEM

(Field emission scanning electron microscopy)



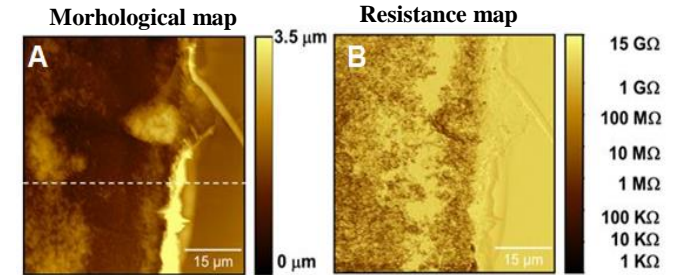
### Raman spectroscopy

CBNPs (Red) CBPTE-5 surface (black).

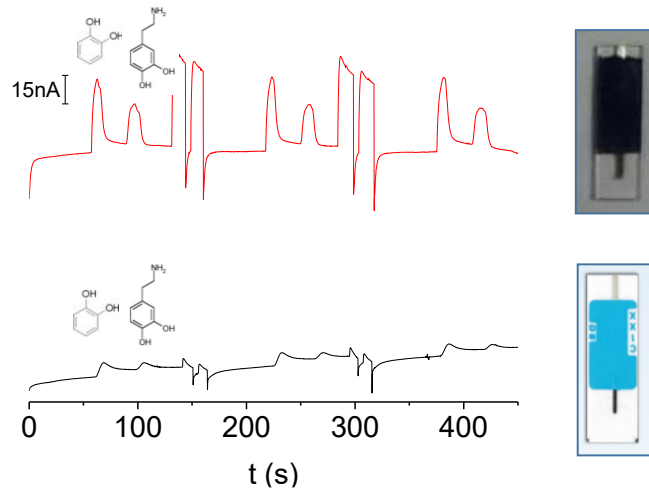
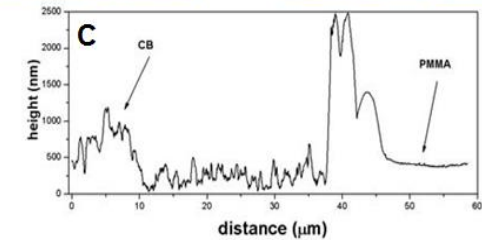


### AFM at the interface

(contact mode)

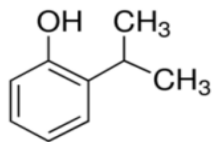


Dashed line

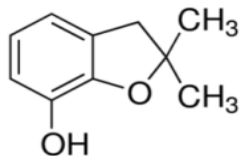


**Detection potential +0.40 V !!**

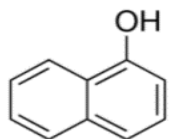
# Microfluidic LAB on CHIP strategy



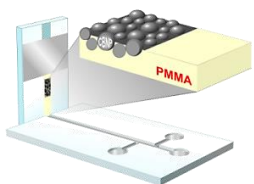
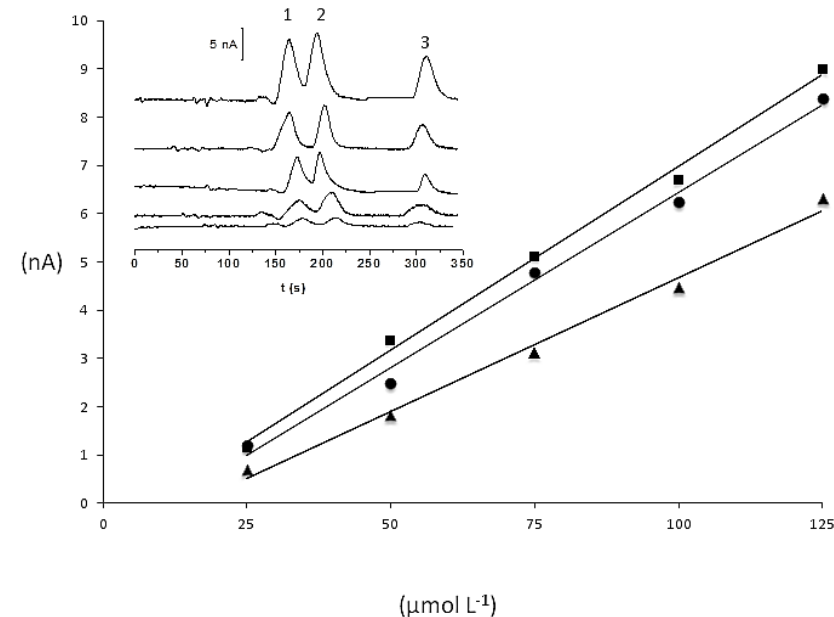
1 peak: Isprocarb  
(2-isopropylphenol)



2 peak: Carbofuran  
(2,3-dihydro-2,2-dimethyl-7-hydroxy benzofuran)



3 peak: Carbaryl  
(1-naphthol)



Analyte	Linear range (μmol L <sup>-1</sup> )	Regression equation (Y=am + b)	Coefficient of determination (r <sup>2</sup> )	Detection limit (μmol L <sup>-1</sup> )	Quantification limit (μmol L <sup>-1</sup> )
<b>Isprocarb (1)</b>	25-125	Y = 0.0726x - 0.830	<b>0.993</b>	26	32
<b>Carbofuran (2)</b>	25-125	Y = 0.0760x - 0.632	<b>0.996</b>	22	28
<b>Carbaryl (3)</b>	25-125	Y = 0.0555x - 0.876	<b>0.992</b>	34	32

# Microfluidic LAB on CHIP strategy

## CBNPs Press-transferred microfluidic carbamates real samples analysis



Pesticide	Lake water <sup>1</sup>				River water <sup>1</sup>				Irrigation water <sup>1</sup>			
	Spiked (μM)	Determined (μM)	Recovery (mean)	RSD (%)	Spiked (μM)	Determined (μM)	Recovery (mean)	RSD (%)	Spiked (μM)	Determined (μM)	Recovery (mean)	RSD (%)
<b>Isoprocab</b>	0	< LOQ	-	-	0	< LOQ	-	-	0	< LOQ	-	-
	50	44.28	<b>87</b>	8	50	52.62	<b>105</b>	8	50	44.97	<b>90</b>	8
	100	90.43	<b>90</b>	5	100	95.94	<b>96</b>	10	100	94.15	<b>94</b>	5
<b>Carbofuran</b>	0	< LOQ	-	-	0	< LOQ	-	-	0	< LOQ	-	-
	50	45.87	<b>92</b>	11	50	53.04	<b>106</b>	9	50	52.97	<b>106</b>	11
	100	89.09	<b>89</b>	9	100	90.61	<b>91</b>	8	100	97.18	<b>97</b>	9
<b>Carbaryl</b>	0	< LOQ	-	-	0	< LOQ	-	-	0	< LOQ	-	-
	50	44.74	<b>90</b>	11	50	45.73	<b>92</b>	9	50	44.83	<b>90</b>	10
	100	88.88	<b>89</b>	9	100	87.53	<b>88</b>	7	100	88.79	<b>89</b>	8

<sup>1</sup> Tests carried out in triplicate.

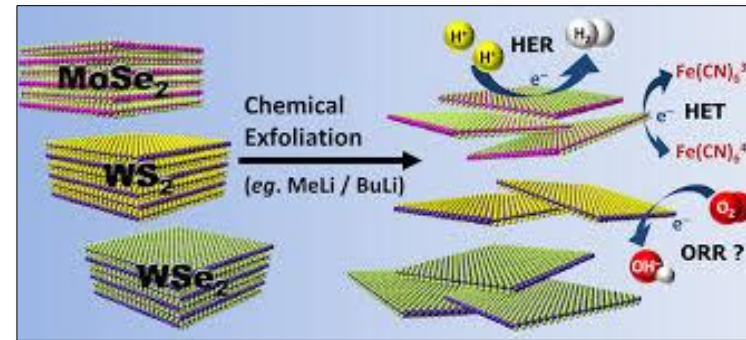
# Graphene-like nanomaterials

## Graphene-like two-dimensional layered nanomaterials, transition metal dichalcogenides (TMDs)

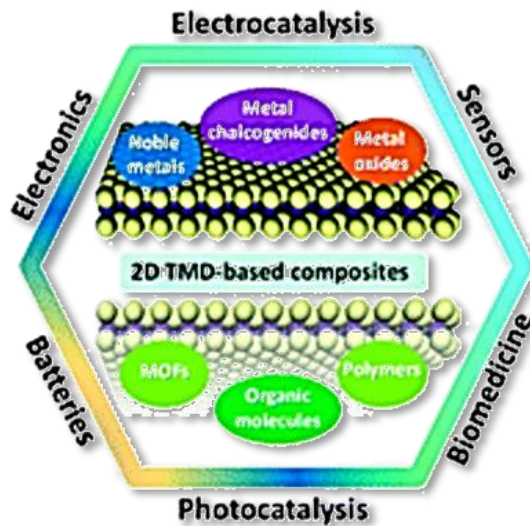
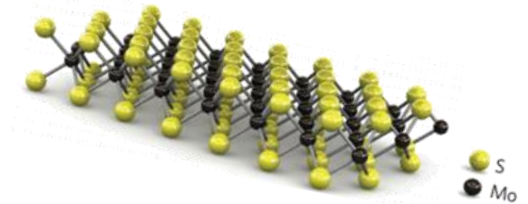
**$MX_2$**  ----- **X = Chalcogen**

**M = Transition Metal**

H																	He
Li	Be											B	C	N	O	F	Ne
Na	Mg	3	4	5	6	7	8	9	10	11	12	Al	Si	P	S	Cl	Ar
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
Cs	Ba	La-Lu	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
Fr	Ra	Ac-Lr	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Uut	Fl	Uup	Lv	Uus	Uuo



Eng, A. Y. S., Ambrosi, A., Sofer, Z., Simek, P., & Pumera, M. (2014). Electrochemistry of transition metal dichalcogenides: strong dependence on the metal-to-chalcogen composition and exfoliation method. *ACS nano*, 8(12), 12185-12198.



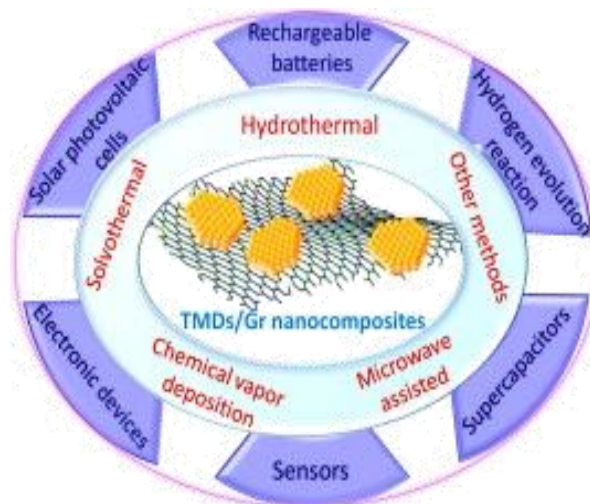
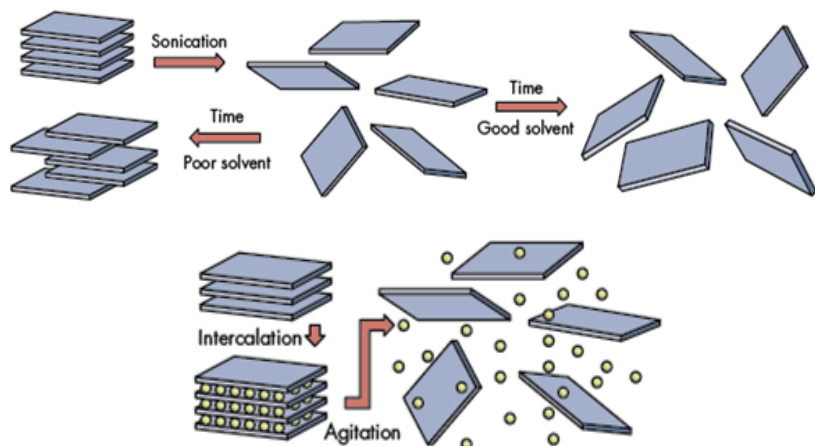
- TMDs nanosheets Easy to prepare
- Unique electrical, optical, and mechanical properties
- Large surface area, low cost, stability
- Metallic and semi-conducting electrical capabilities
- Widely employed in hydrogen evolution reaction (HER) and energy storage
- Few applications in (bio)sensors employed in food analysis
- Tunable electrocatalytic properties /intercalatable morphologies



## (2) Graphene-like nanomaterials

### TMDs and TMDs-based nanocomposite for sensor and biosensors

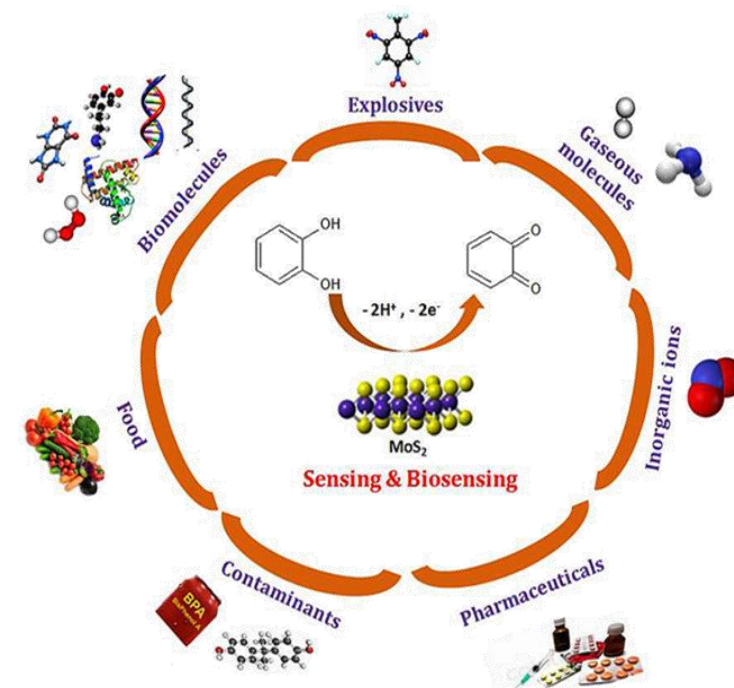
2D NMs can form heterostructures with layers of varied materials and with a thickness of one or two atoms, and thus synergistically improve their physicochemical properties.



### TMDs-based nanocomposites/ Hybrid nanoarchitectures

These strategies avoiding restacking, narrow potential window, low conductivity, fouling, etc.

**improving the general electrochemical performance**

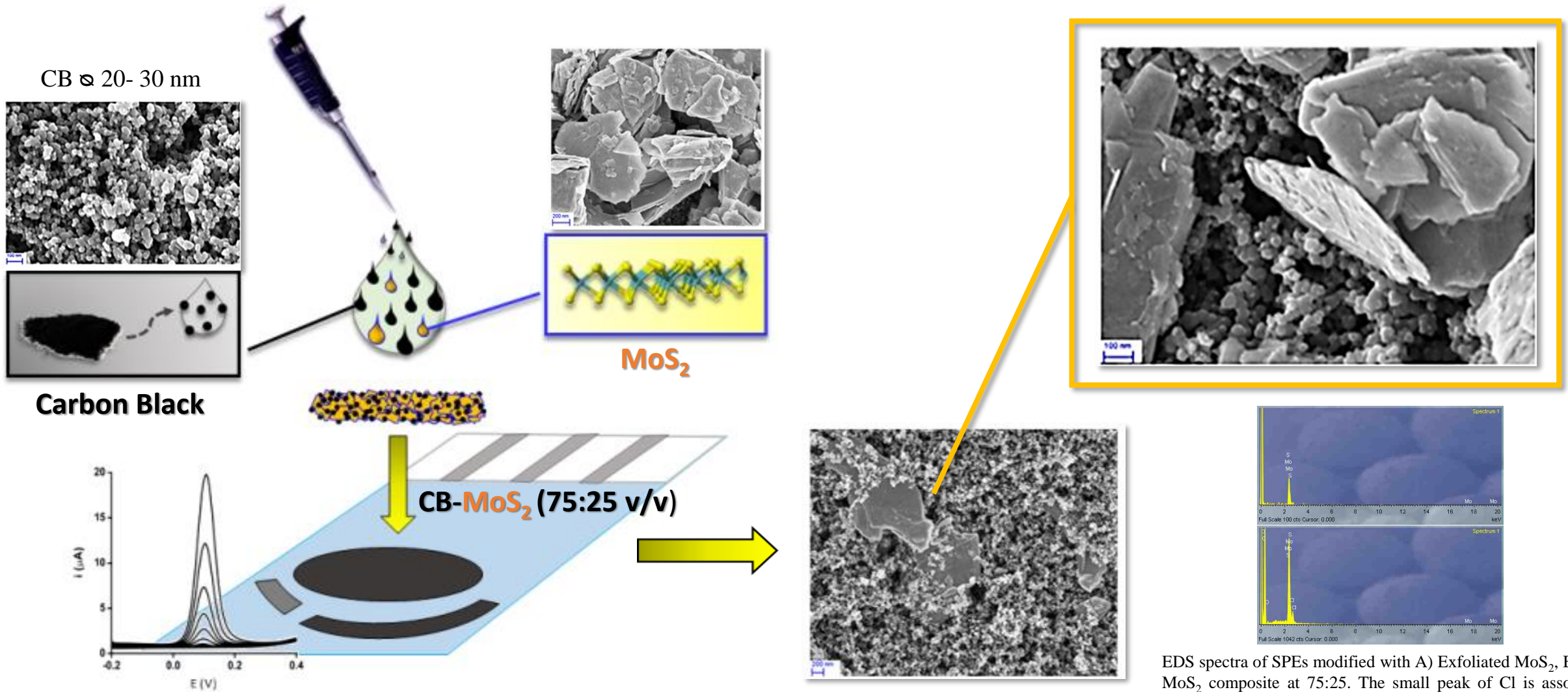


*TMDs-based sensors and biosensors growing field and holds great promise*

**Exfoliation using intercalating materials/  
appropriate solvent**  
(single- or few-layer TMDs nanosheets)

# Graphene-like nanohybrids materials

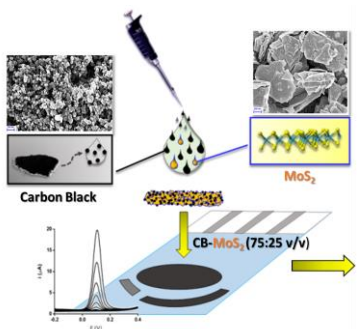
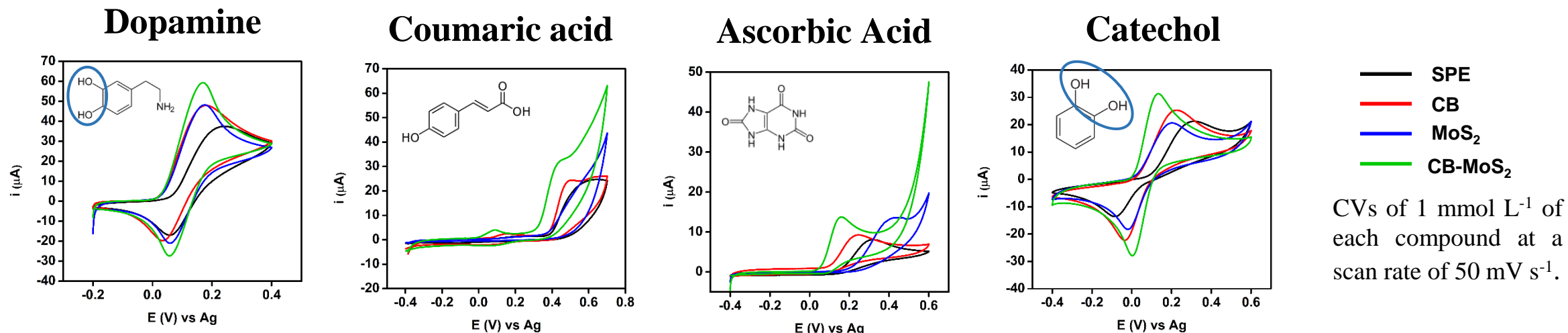
Effective nanocomposite carbon black(CB)-molybdenum disulphide ( $\text{MoS}_2$ ) as novel screen-printed electrodes modifier for sensing applications



EDS spectra of SPEs modified with A) Exfoliated  $\text{MoS}_2$ , B)  $\text{CB-MoS}_2$  composite at 75:25. The small peak of Cl is associated with external contamination.

# Graphene-like nanohybrids materials

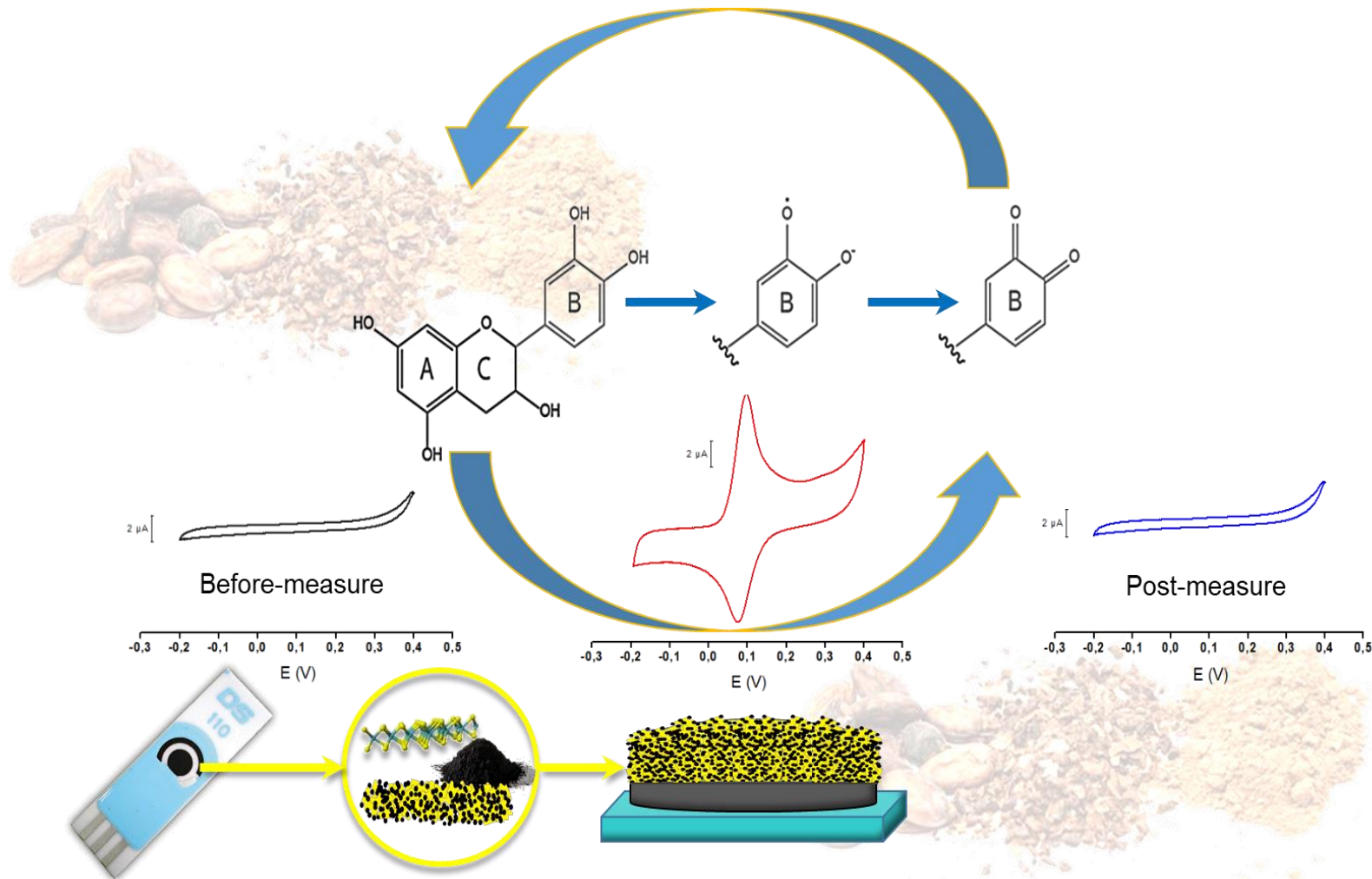
Electrochemical response towards common electroactive species



- Dopamine and Catechol showed an improvement in the peak-to-peak separation and increased peak intensities.
- Coumaric acid and ascorbic acid showed a negative shift and improved peak intensities.
- SPE-CB-MoS<sub>2</sub> anodic peak intensity decrease was in the  $\leq 2\%$  and  $\leq 10\%$  for Catechol/Dopamine and Uric acid/Coumaric acid, respectively, after 10 scans

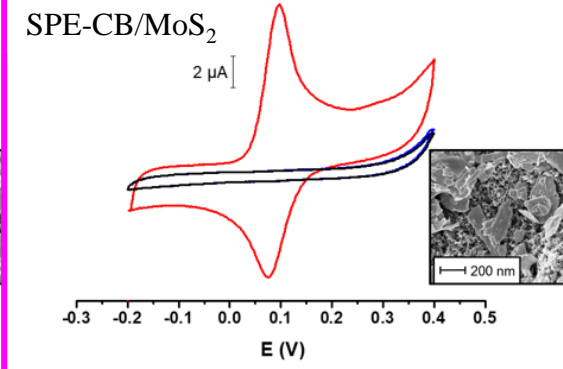
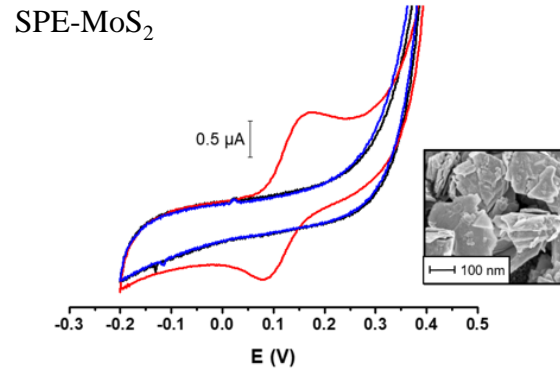
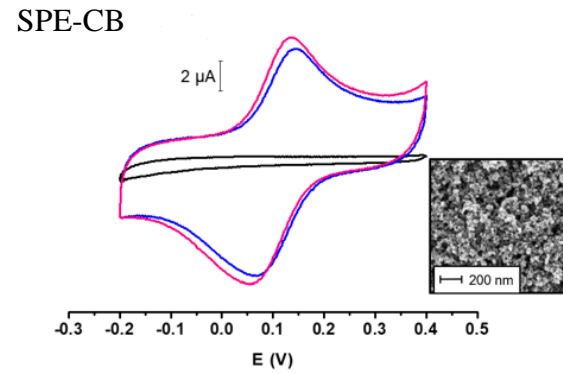
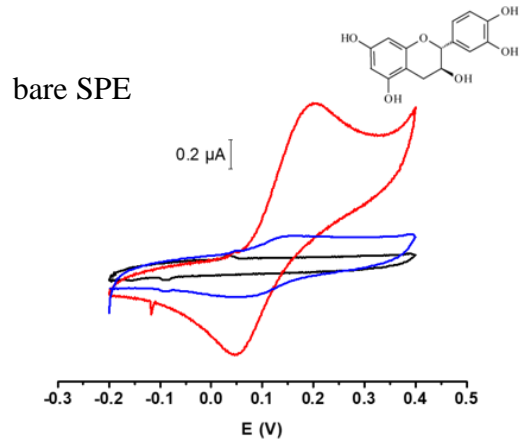
# (6) Graphene-like nanohybrids materials

## A case of study: Cocoa polyphenols



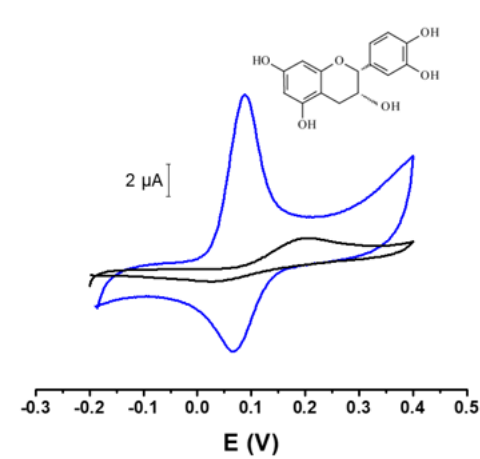
# Graphene-like nanohybrids materials

## CATECHIN

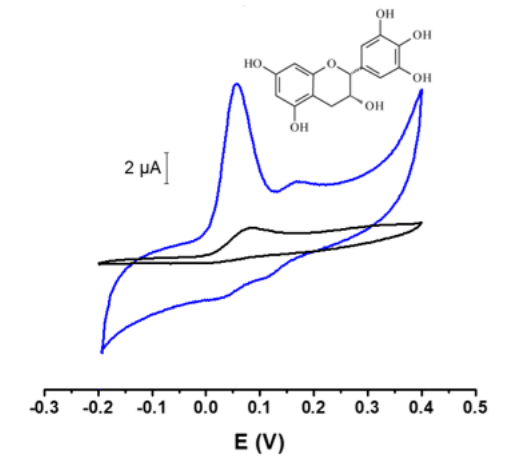


Anti-Fouling properties

## EPICATECHIN

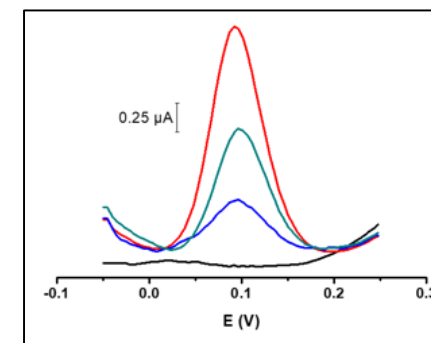
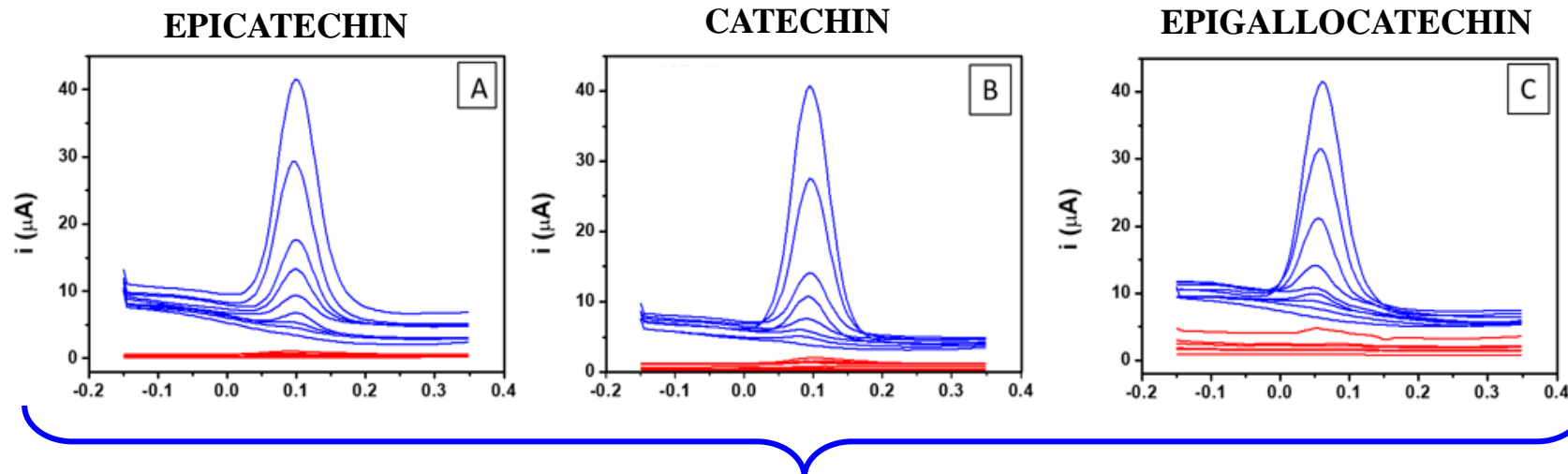


## EPIGALLOCATECHIN



## Standard calibration curve

## Fortified cocoa samples

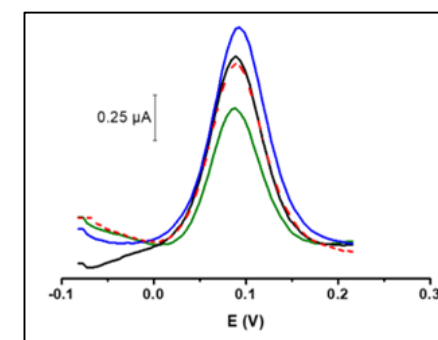


- + 0,25 μM catechin
- + 0,5 μM catechin
- Cocoa mix

Analytical characteristics of the SPE-CB/MoS<sub>2</sub> sensor employed for CT, EP, and EG detection.

	Linear Range (μmol L <sup>-1</sup> )	R <sup>2</sup>	Sensitivity (μA L μmol <sup>-1</sup> )	LOD (μmol L <sup>-1</sup> )
CT	0.2-25	0.998	1.12	0.18
EP	0.2-25	0.998	1.18	0.17
EG	0.2-25	0.998	1.10	0.18

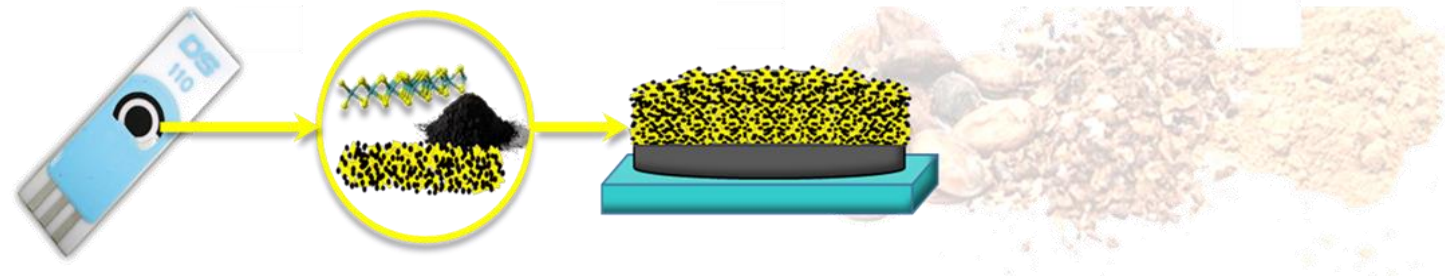
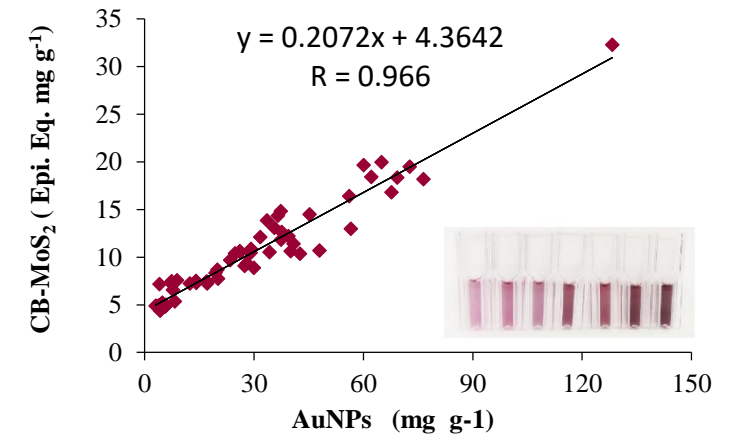
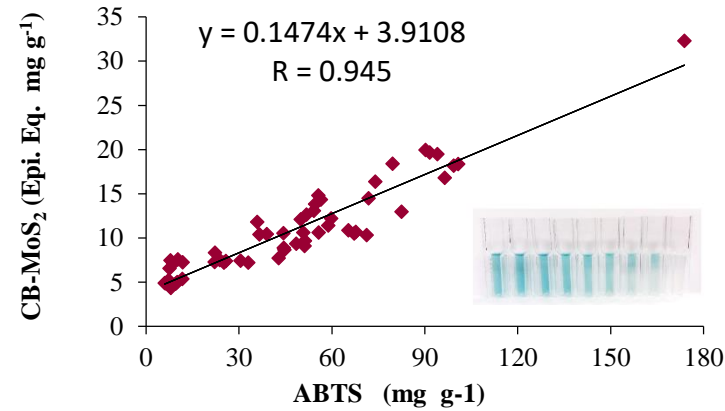
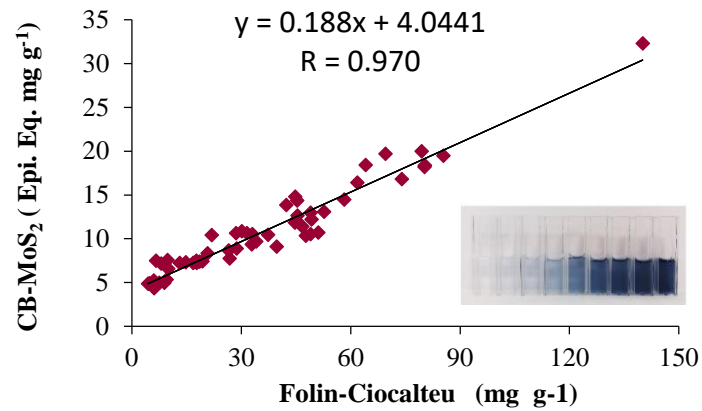
<sup>1</sup>LODs were calculated as  $3\sigma/\text{slope}$  ratio, where  $\sigma$  is the standard deviation of the mean value for 10 voltammograms of the blank. Analytical characteristics calculated using the mean value of three calibration curves.



- Low polyphenols content
- Medium polyphenols content
- High polyphenols content
- - - n= 5 repetition after n° 59 cocoa measurement

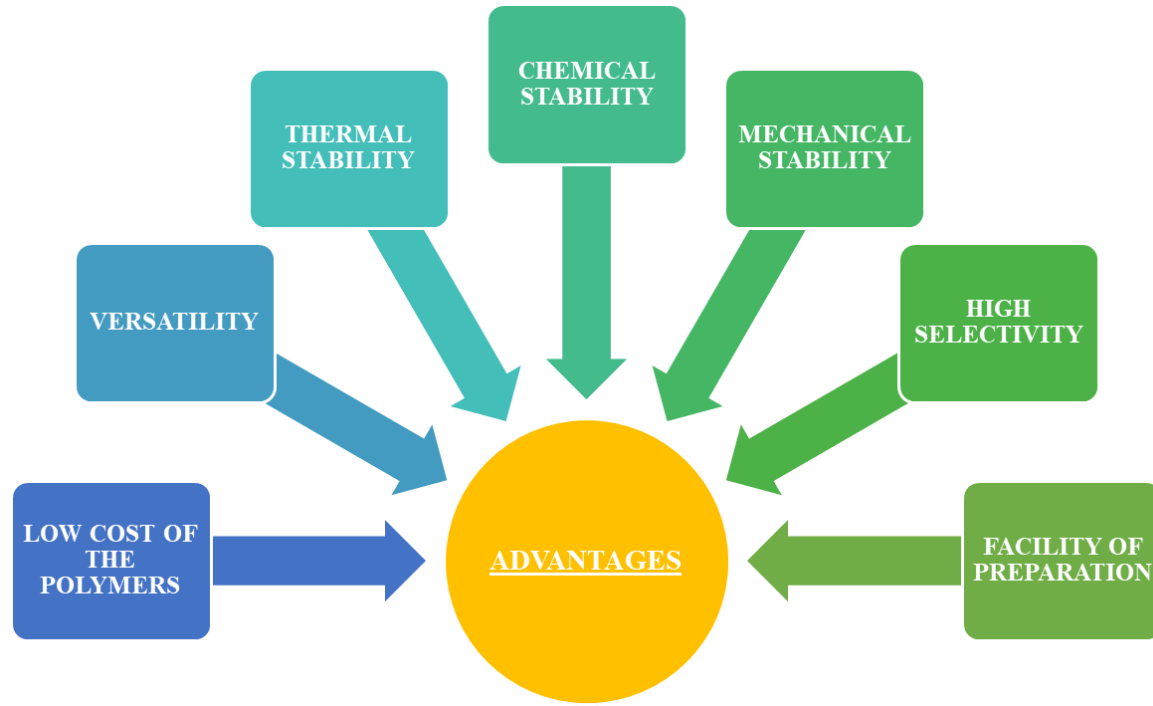
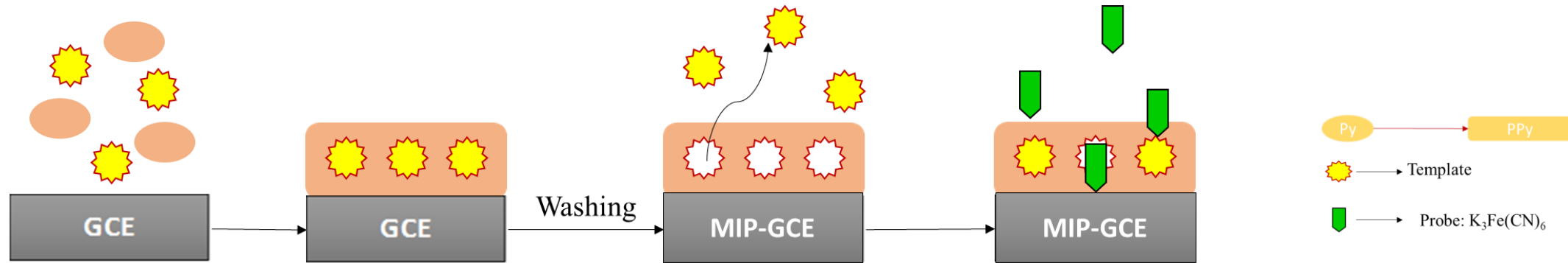


## Data correlation Optical assay vs. Electrochemical n= 59 sample

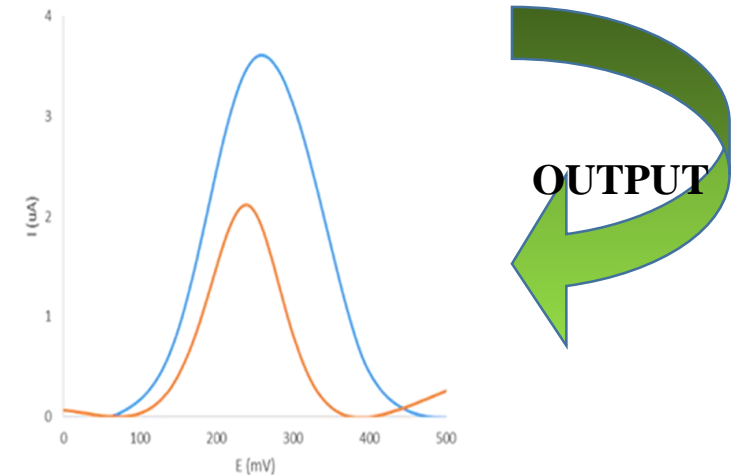


# Molecular imprinted polymer (MIP) based sensing

A **molecularly imprinted polymer (MIP)** is a **polymer** that has been processed using the **molecular imprinting** technique which leaves cavities in the polymer matrix with an affinity for a chosen "template" molecule.



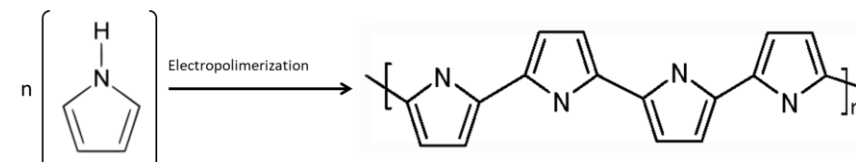
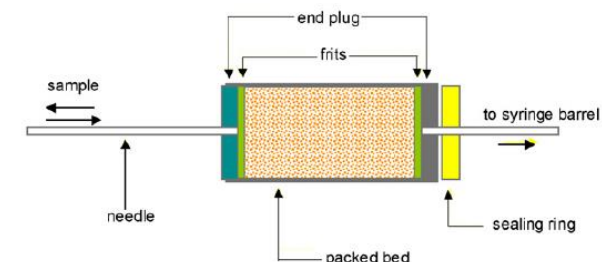
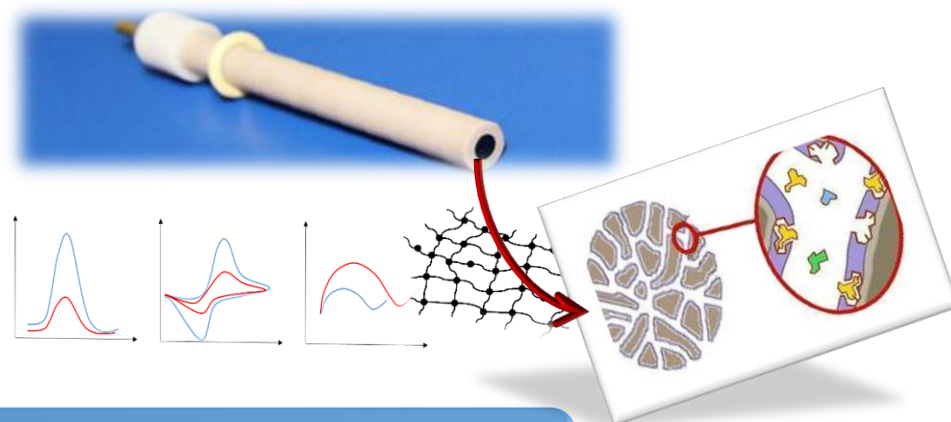
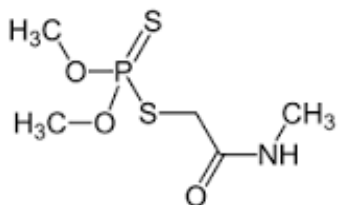
## REBINDING & MEASURE





# (1) MIP based sensing

**MIP-MEPS based sensing strategy for the selective assay of dimethoate. Application to wheat flour samples**



**DIMETHOATE MONITORING  
IN WHEAT FLOUR**

**SAMPLE  
PREPARATION**

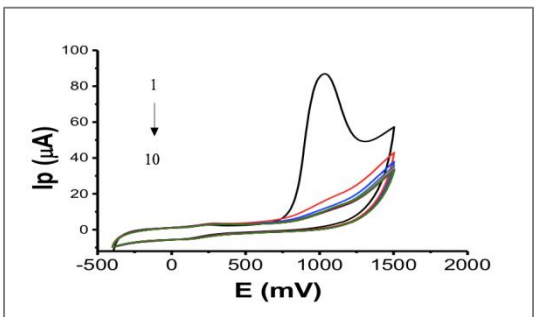
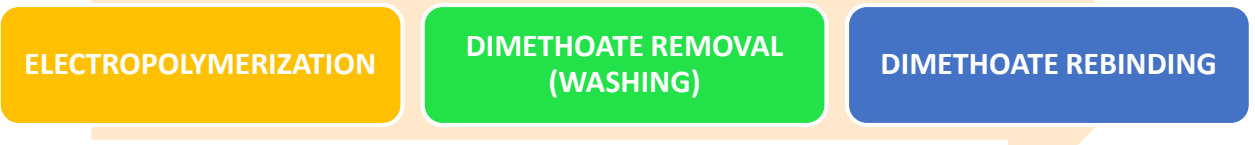
**ANALYTE  
DETECTION**

**MICROEXTRACTION  
BY PACKED  
SORBENT (MEPS)**

**MIP-GLASSY  
CARBON  
ELECTRODE**

# (2) MIP based sensing

## Strategy



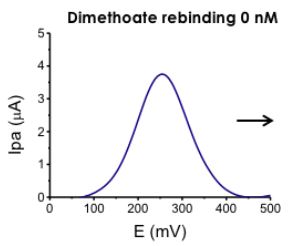
- Dimethoate (dim)
- Pyrrole (Py)

### ELECTROPOLYMERIZATION

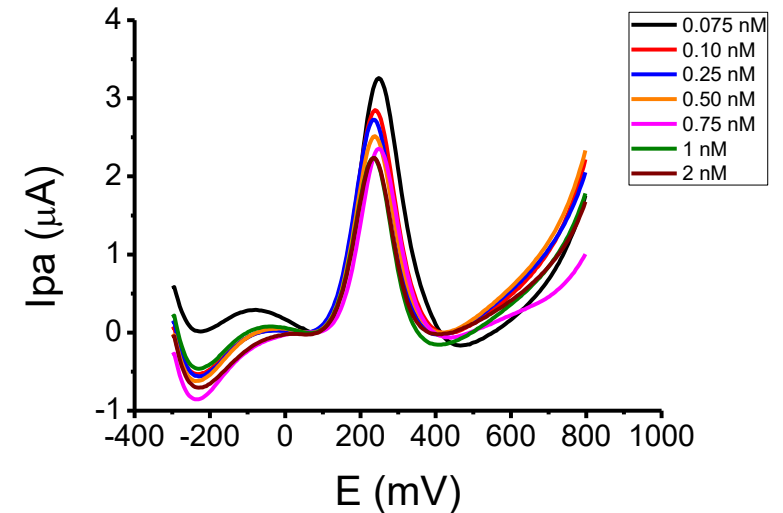
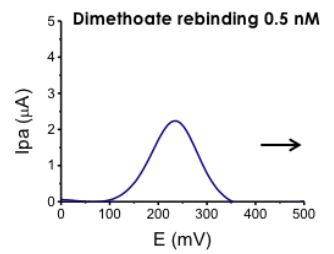


Probe  $K_3[Fe(CN)_6]$   
10 mM

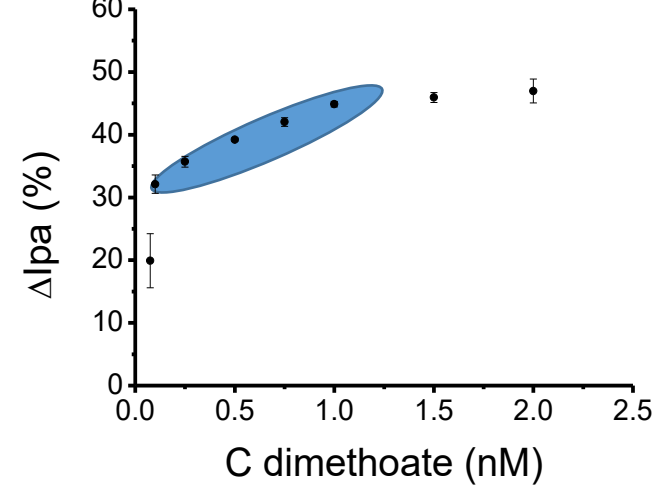
### WASHING in HCl solution



### REBINDING (15 min)



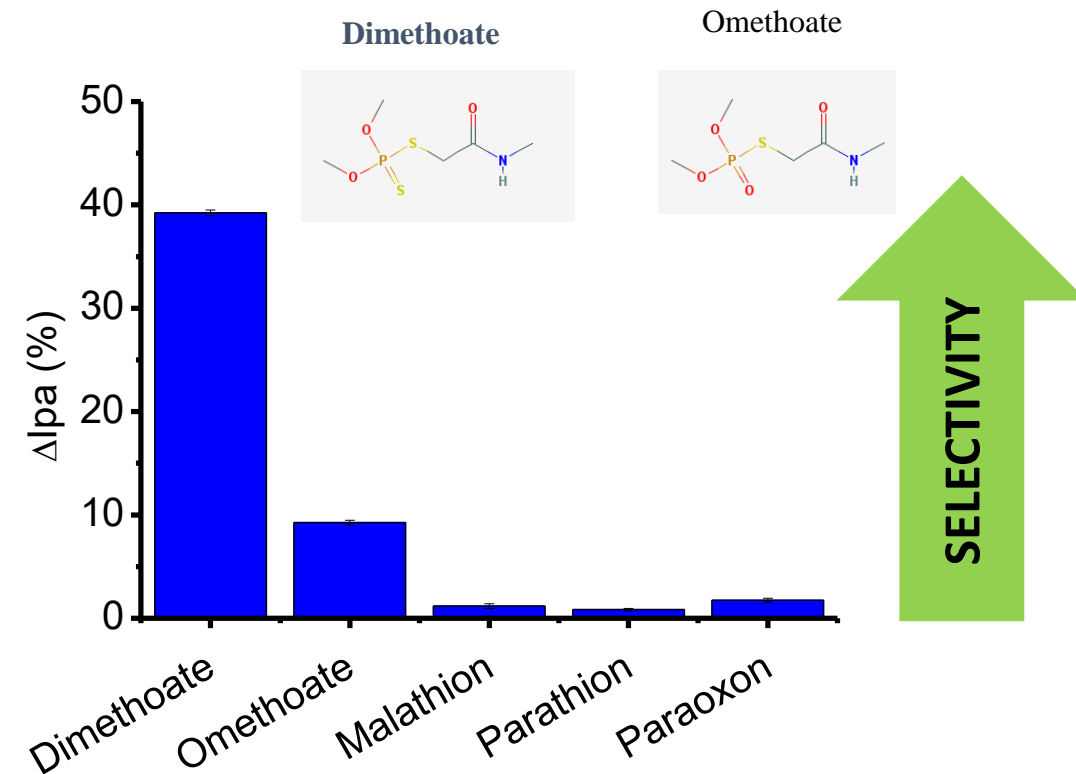
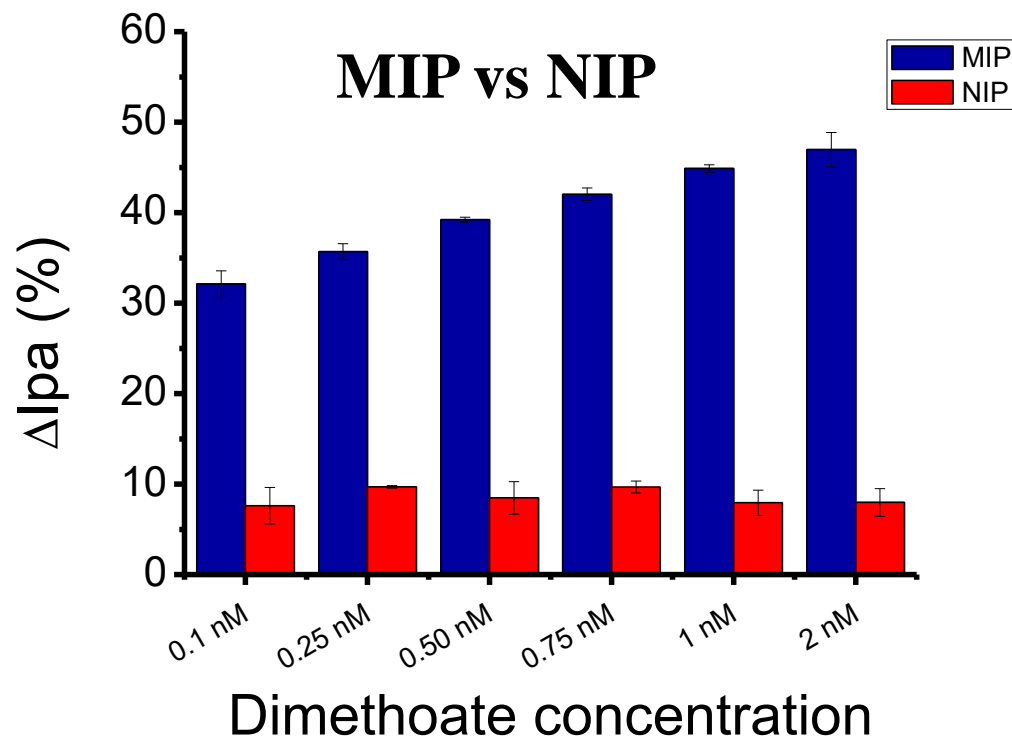
### Dose-response curve



### (3) MIP based sensing

$\Delta I_{pa}$ (%)	Repeatability (RSD %)	Reproducibility (RSD %)
0.5 nM dimethoate (n=3)	0.7	2.7
1 nM dimethoate (n=3)	0.9	5.5

$\Delta I_{pa}$  (%) for malathion, parathion and paraoxon after the rebinding step was negligible; **omethoate** gave a response of **23%**.



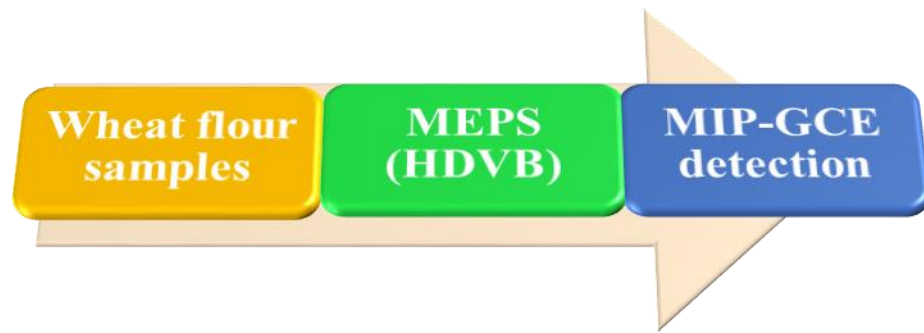
# (4) MIP based sensing



MIP-MEPS based sensing strategy for the selective assay of dimethoate. Application to wheat flour samples

D. Capoferri<sup>a</sup>, M. Del Carlo<sup>a,1</sup>, N. Ntshongontshi<sup>b</sup>, E.I. Iwuoha<sup>b</sup>, M. Sergi<sup>a</sup>, F. Di Ottavio<sup>a</sup>, D. Compagnone<sup>a,\*,1</sup>

<sup>a</sup> Faculty of Biosciences and Technologies for Food, Agriculture and Environment, University of Teramo, via R. Balzarini 1, 64100 Teramo, Italy  
<sup>b</sup> SensorLab, Department of Chemistry, University of the Western Cape, Bellville 7535, South Africa



## Wheat flour samples: MIP vs. UHPLC-MS/MS

### Samples

MIP-GCE  
RELATIVE ERROR (%) of dimethoate  
concentration ( $\mu\text{g kg}^{-1}$ )

MIP-GCE  
SD of dimethoate  
concentration ( $\mu\text{g kg}^{-1}$ )

- Wheat flour spiked with dimethoate 0.5 MRL
- Wheat flour spiked with dimethoate 0.5 MRL + mix
- Wheat flour spiked with dimethoate MRL
- Wheat flour spiked with dimethoate MRL + mix
- Wheat flour spiked with dimethoate 1.5 MRL
- Wheat flour spiked with dimethoate 1.5 MRL + mix
- Wheat flour spiked with dimethoate MRL + omethoate (1:1)

+13.5

0.52

+4.6

2.37

-21.1

1.24

-21.2

1.36

+16.7

0.74

-0.4

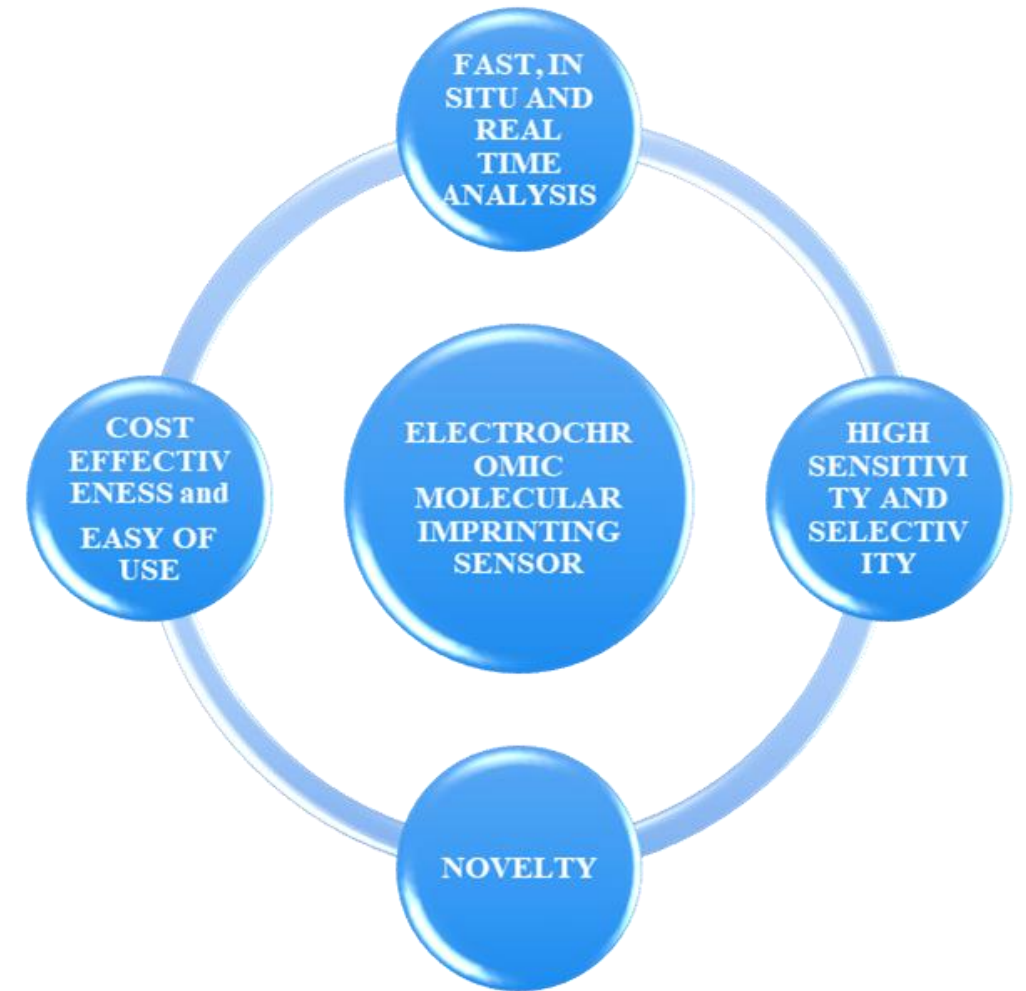
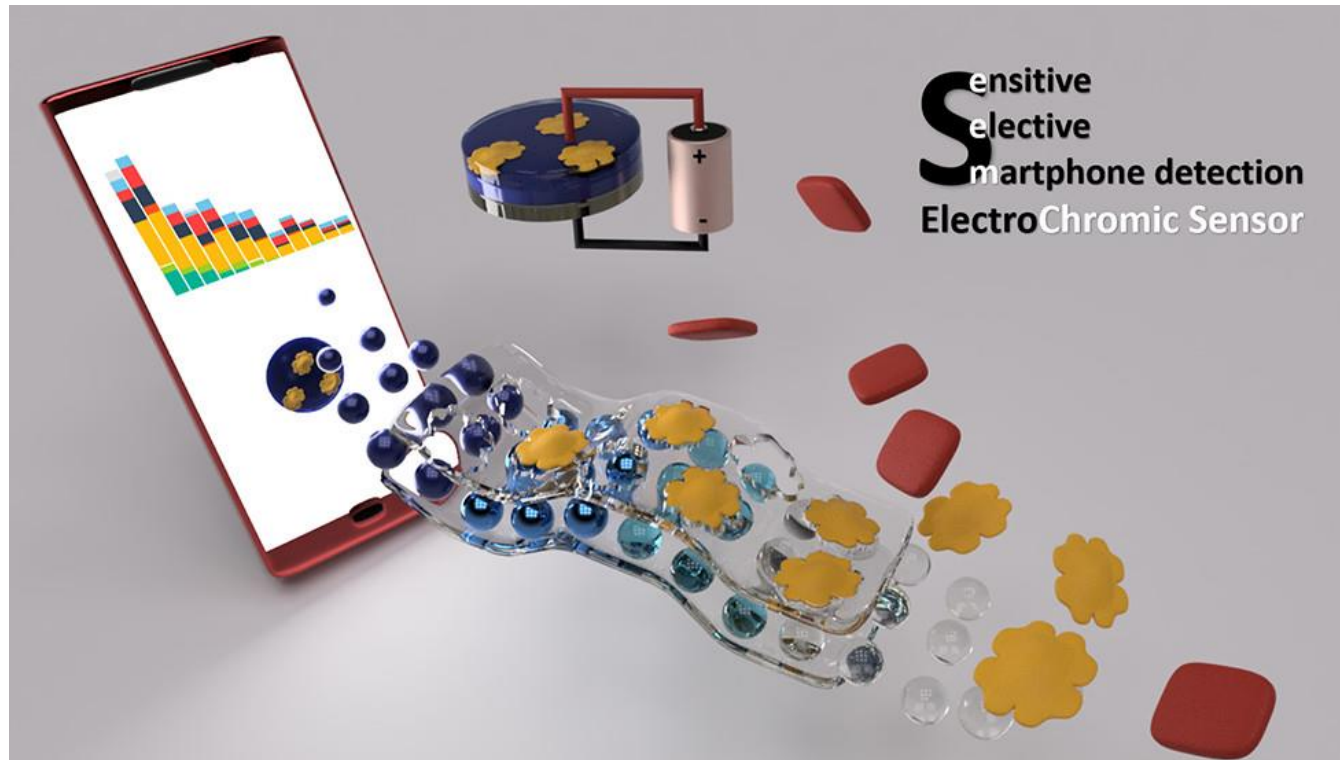
1.69

+3.5

2.70

## (5) MIP based sensing

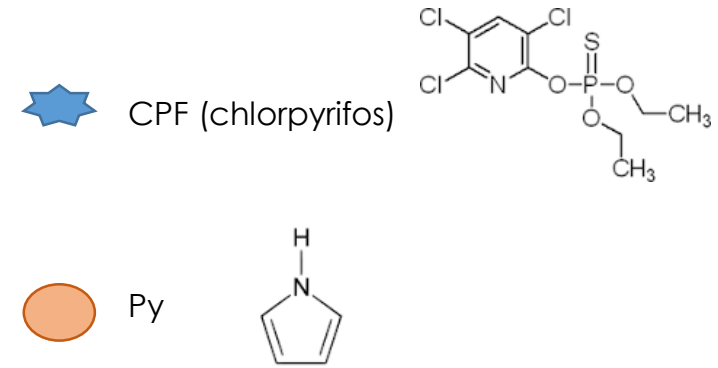
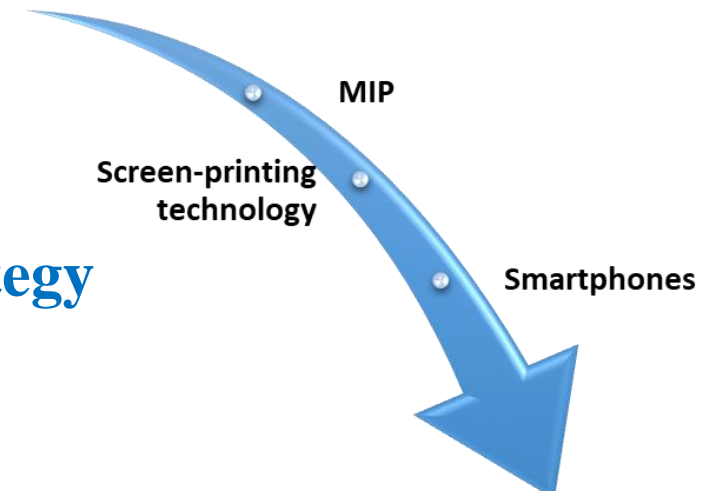
### Electrochromic Molecular Imprinting Sensor for Visual and Smartphone-Based Detections



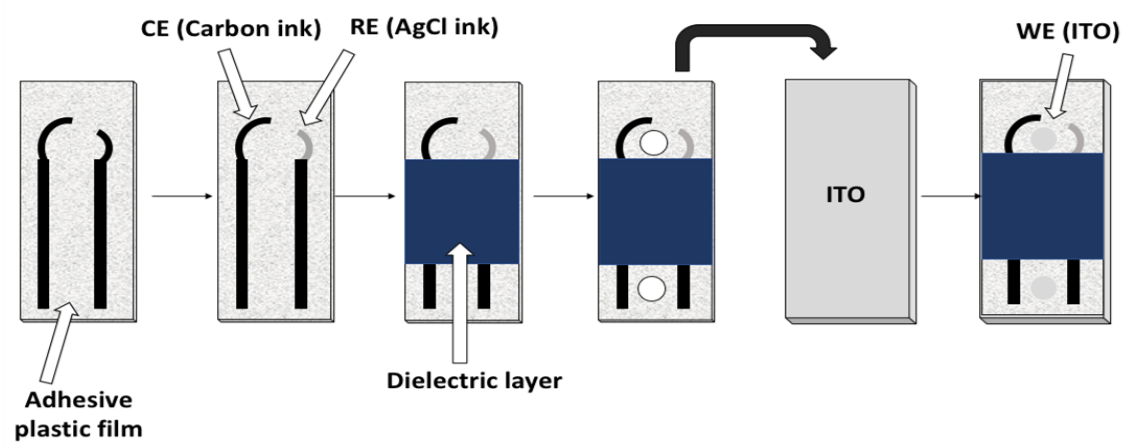
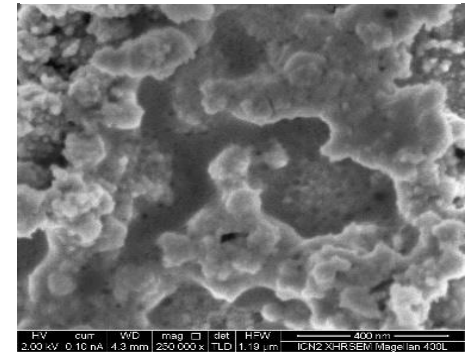
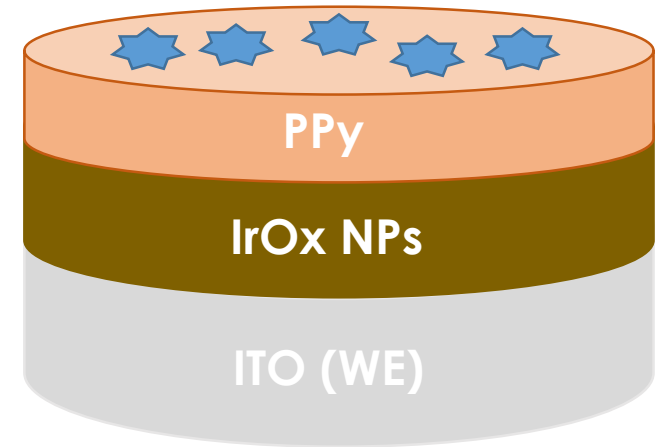
# (6) MIP based sensing

Strategy

Electrochromism

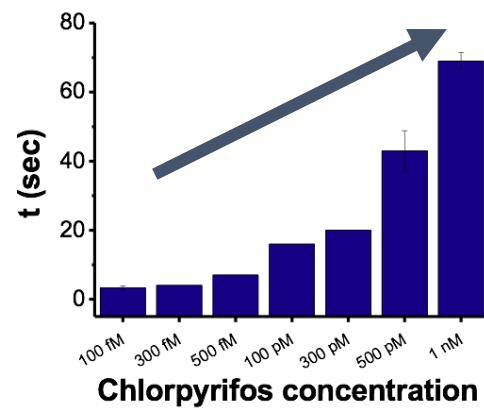
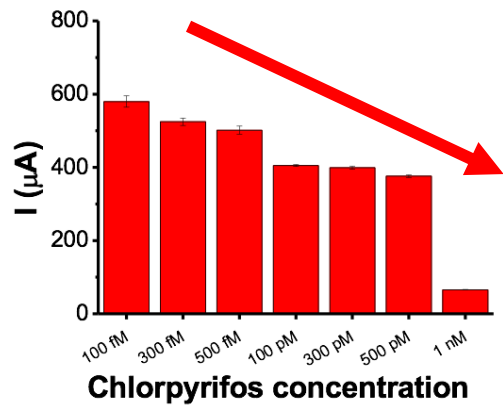
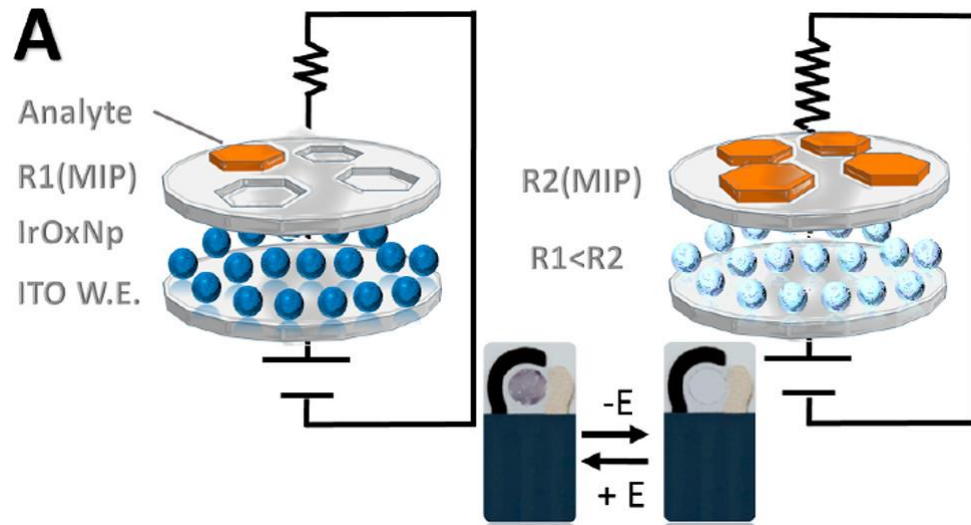


MIP/IrOx NPs – ITO SPE



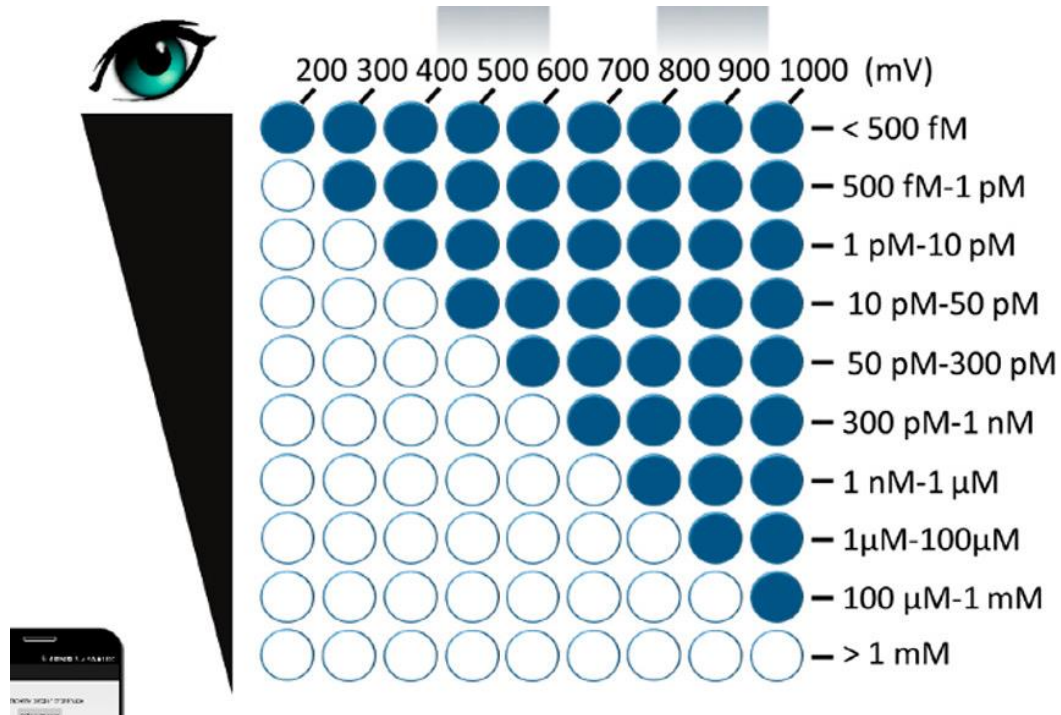
# (7) MIP based sensing

## WORKING PRINCIPLE

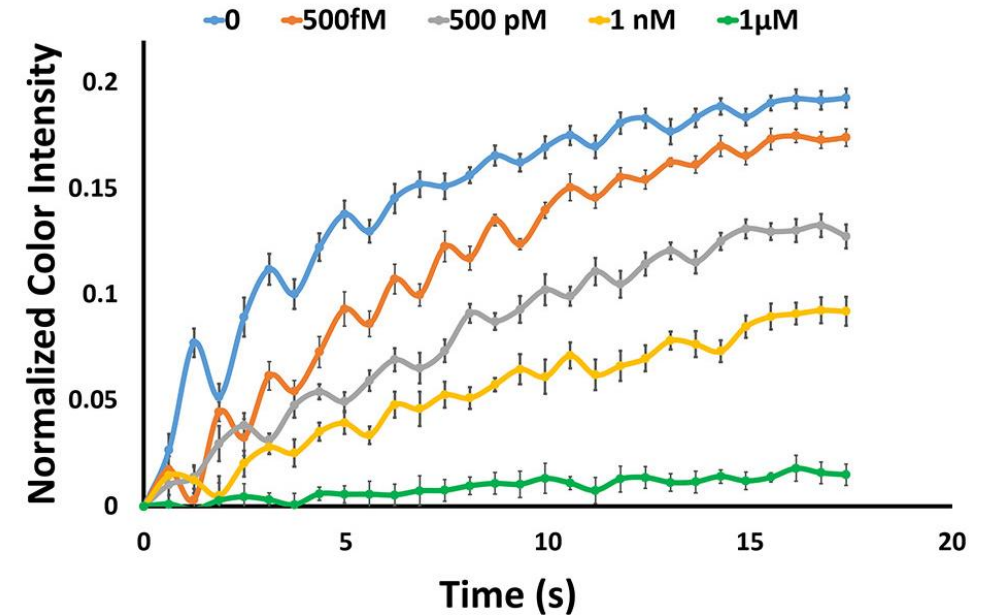


# (8) MIP based sensing

## VISUAL APPROACH



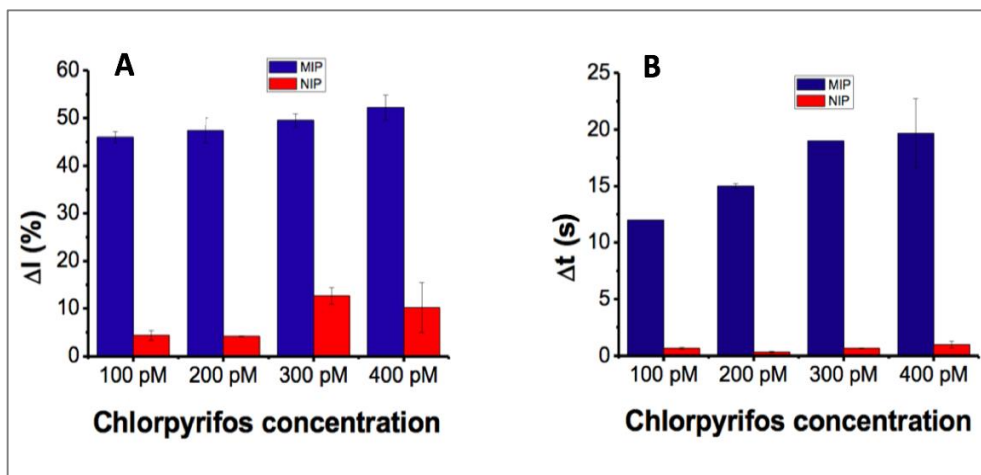
## SMARTPHONE APPROACH



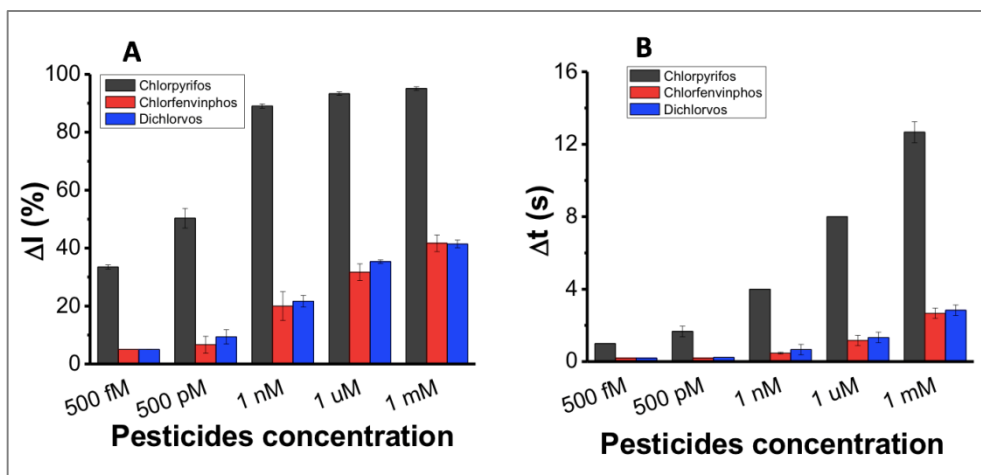


# (9) MIP based sensing

## MIP vs NIP



## SELECTIVITY (500 mV-1000 mV)



## Electrochromic Molecular Imprinting Sensor for Visual and Smartphone-Based Detections

Denise Capoferri,<sup>†,‡,§</sup> Ruslan Álvarez-Diduk,<sup>†,§</sup> Michele Del Carlo,<sup>‡</sup> Dario Compagnone,<sup>‡</sup> and Arben Merkoçi<sup>\*,†,||</sup>

<sup>†</sup>Nanobioelectronics and Biosensor Group, Catalan Institute of Nanoscience and Nanotechnology (ICN2), CSIC, The Barcelona Institute of Science and Technology, Campus UAB, Bellaterra, 08193, Barcelona, Spain

<sup>‡</sup>Faculty of Biosciences and Technologies for Food, Agriculture and Environment, University of Teramo, via R. Balzarini 1, 64100 Teramo, Italy

<sup>§</sup>Catalan Institution for Research and Advanced Studies (ICREA), Pg. Lluís Companys 23, 08010 Barcelona, Spain



## Recovery values of chlorpyrifos in spiked drinking water samples (n = 3) using the current response

Added (Spiked)	Found	Recovery (%)	RSD (%)
500 fM	517.19 fM	103.44 ± 16.14	15.60
500 pM	471.45 pM	94.29 ± 17.92	19.00
1 nM	0.99 nM	99.50 ± 19.90	20.00
1 μM	0.98 μM	97.55 ± 25.87	26.52
1 mM	1.07 mM	106.57 ± 15.30	14.36



Review

## Affinity Sensing Strategies for the Detection of Pesticides in Food

Denise Capoferri, Flavio Della Pelle , Michele Del Carlo and Dario Compagnone \* 

Faculty of Biosciences and Technologies for Food, Agriculture and Environment, University of Teramo, via R. Balzarini 1, 64100 Teramo, Italy; dcapoferri@unite.it (D.C.); fdellapelle@unite.it (F.D.P.); mdelcarlo@unite.it (M.D.C.)



\* Correspondence: dcompagnone@unite.it; Tel.: +39-0861-266942

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Review

## Nanomaterial-Based Sensing and Biosensing of Phenolic Compounds and Related Antioxidant Capacity in Food

Flavio Della Pelle , and Dario Compagnone \* 

Faculty of Bioscience and Technology for Food, Agriculture and Environment, University of Teramo, 64023 Teramo, Italy; fdellapelle@unite.it

\* Correspondence: dcompagnone@unite.it; Tel.: +36-0861-266942

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*questions?*

*Doubts?*

*Some ideas?*



*...Thanks for your attention...*